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MICROSCALE METHOD TO DETERMINE EQUIVALENT FLAMMABILITY (SIMILARITY) OF COMBUSTIBLE COMPONENTS



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Overview of Talk

- *Microscale (10^{-6} kg) fire growth capacity (FGC) combines flame spread & burning rate to better compare materials.*
- *Increased accuracy, repeatability and reproducibility of FGC by accounting for baseline drift in MCC (ASTM Ballot Item).*
- *Method of calculating FGC from MCC data.*

$$FGC = \left(\frac{Q_{\infty}}{T_{burn} - T_{ign}} \right) \left(\frac{T_{burn} - T_0}{T_{ign} - T_0} \right)$$

- *FGC of 30 polymers and flammability correlations.*
- *Repeatability of FGC.*

14 CFR 25 (FAA) Fire Tests



FAA Rate of Heat
Release Apparatus
(Aircraft materials)



Radiant Panel

- Flooring
- Insulation
- Aircraft materials

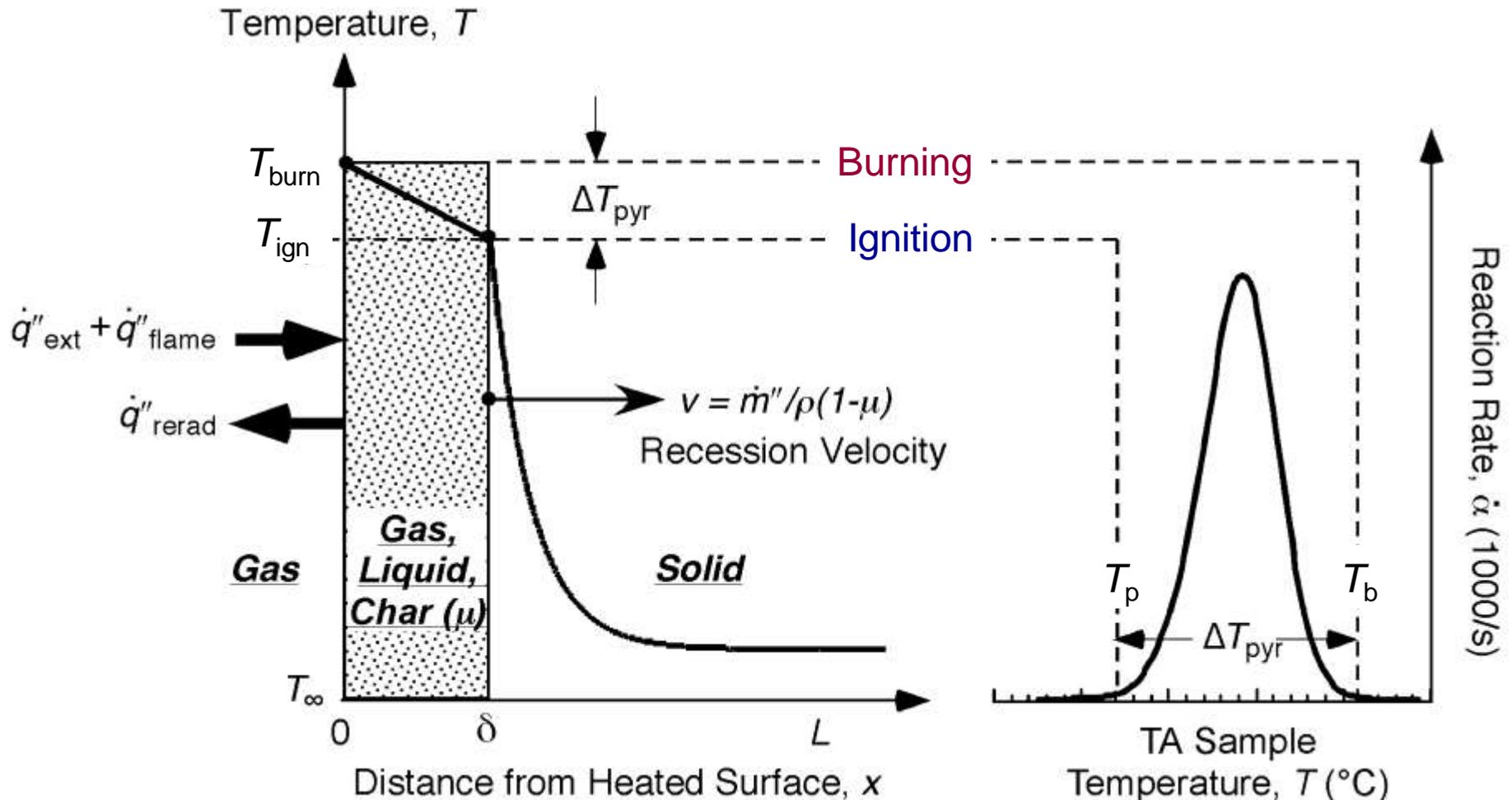


Vertical Bunsen Burner

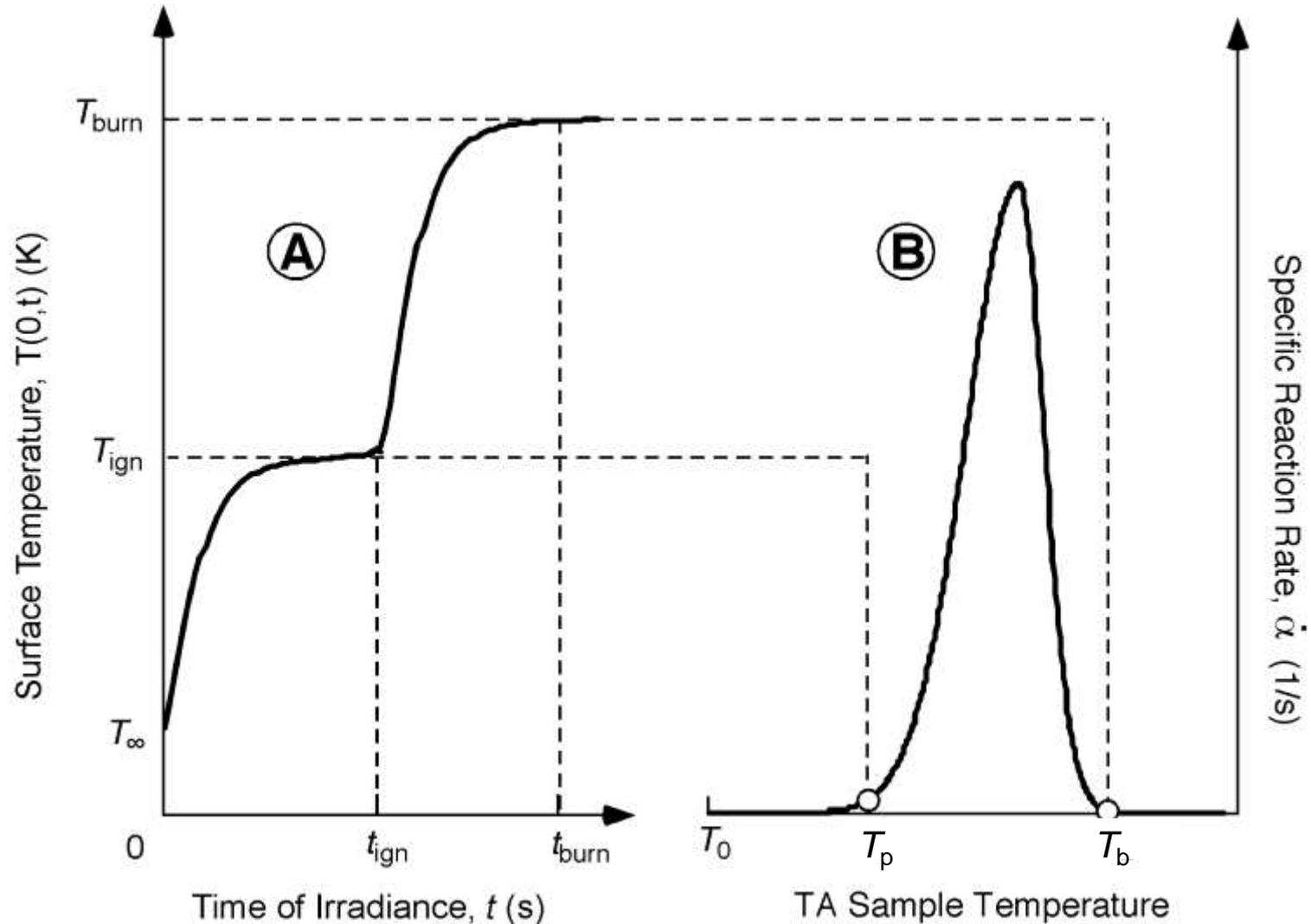
- UL
- FAA
- ASTM

Burning Model of Solids

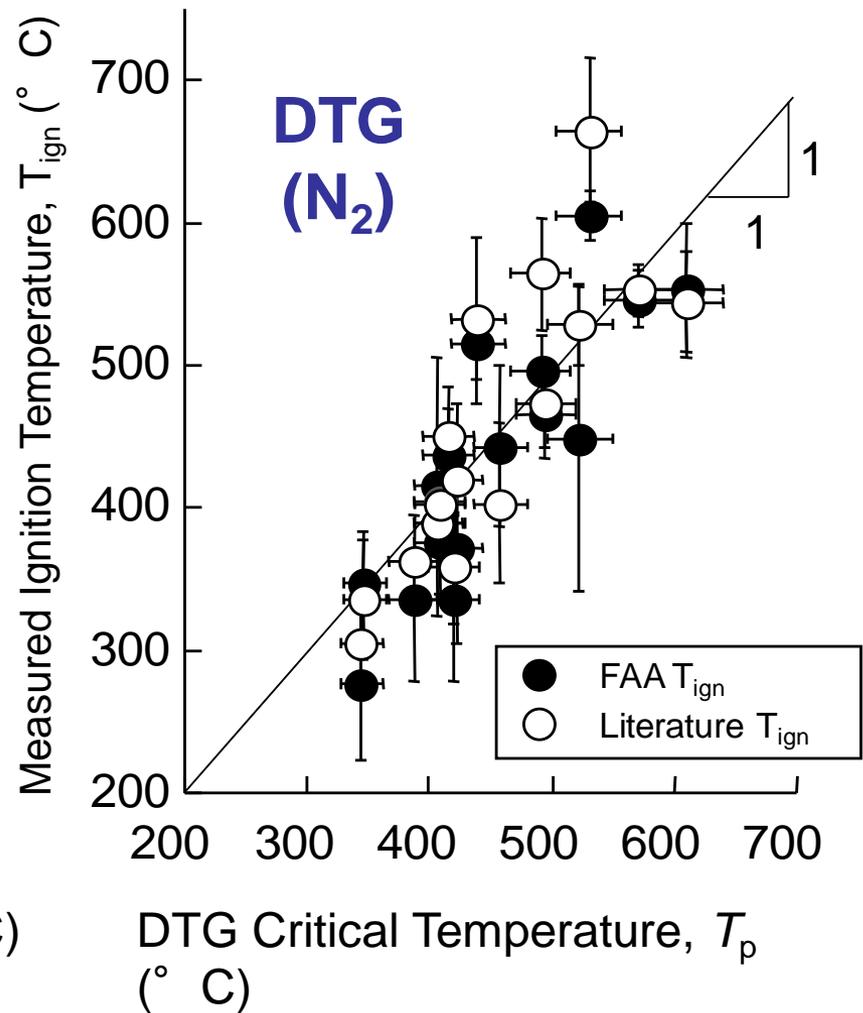
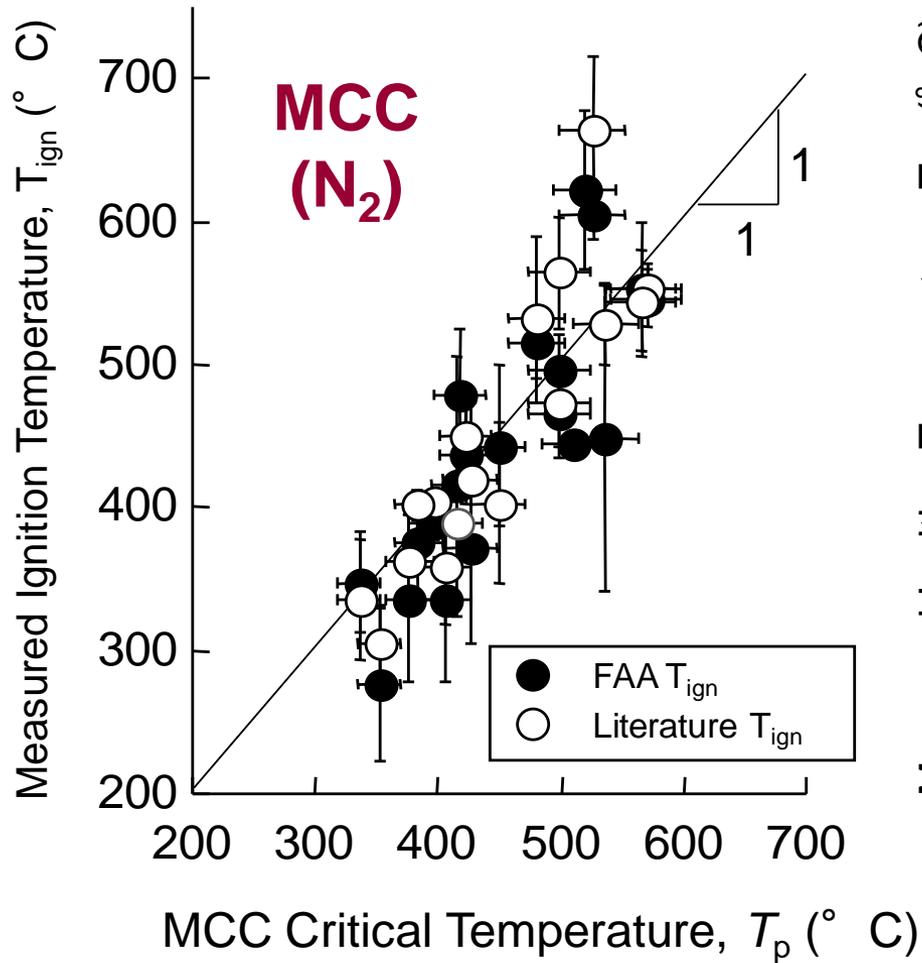
- T_{burn} and T_{ign} bound pyrolysis zone in fire
- T_b and T_p bound pyrolysis interval in MCC



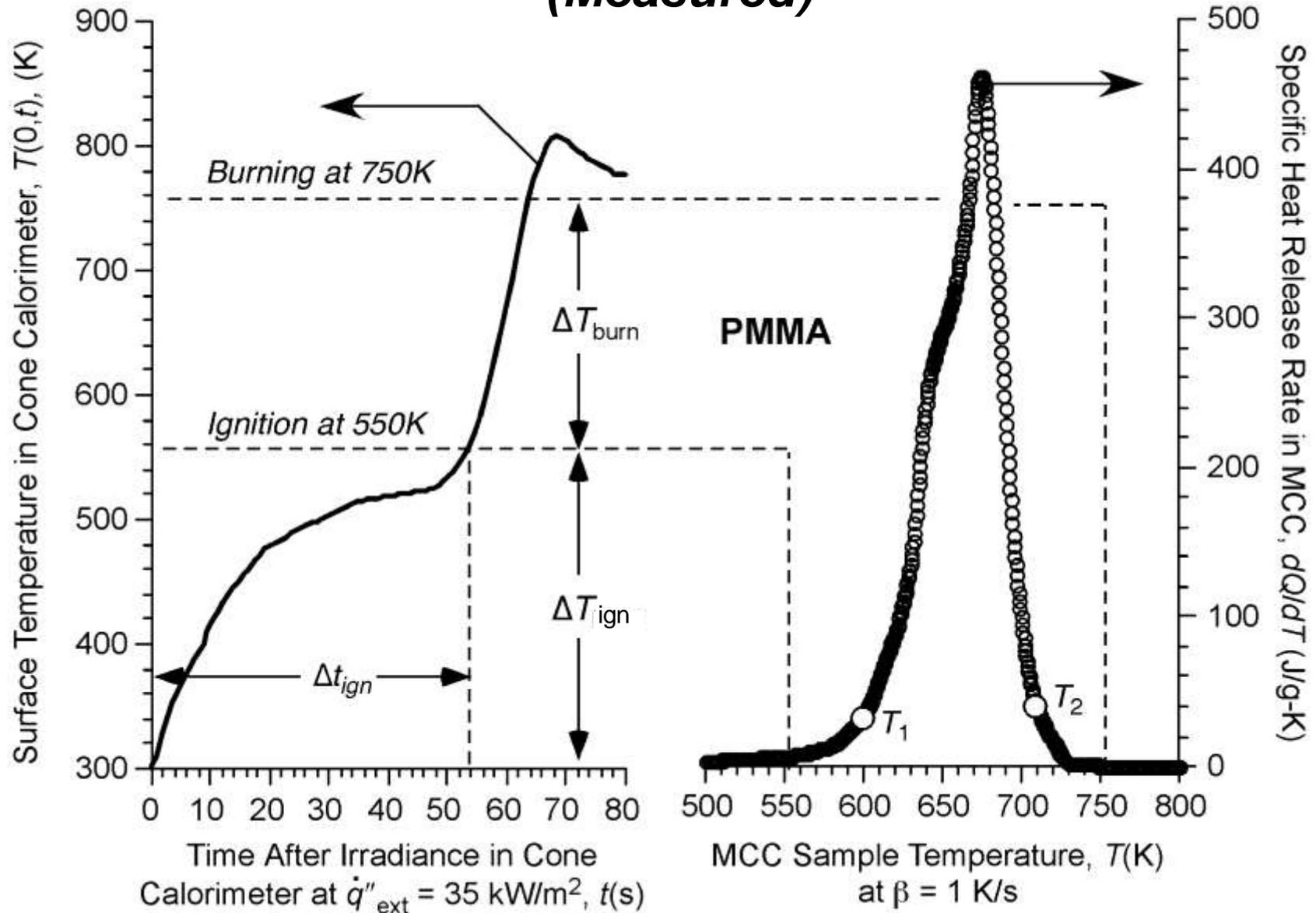
Fire Temperatures in Cone Calorimeter and MCC (Hypothetical)



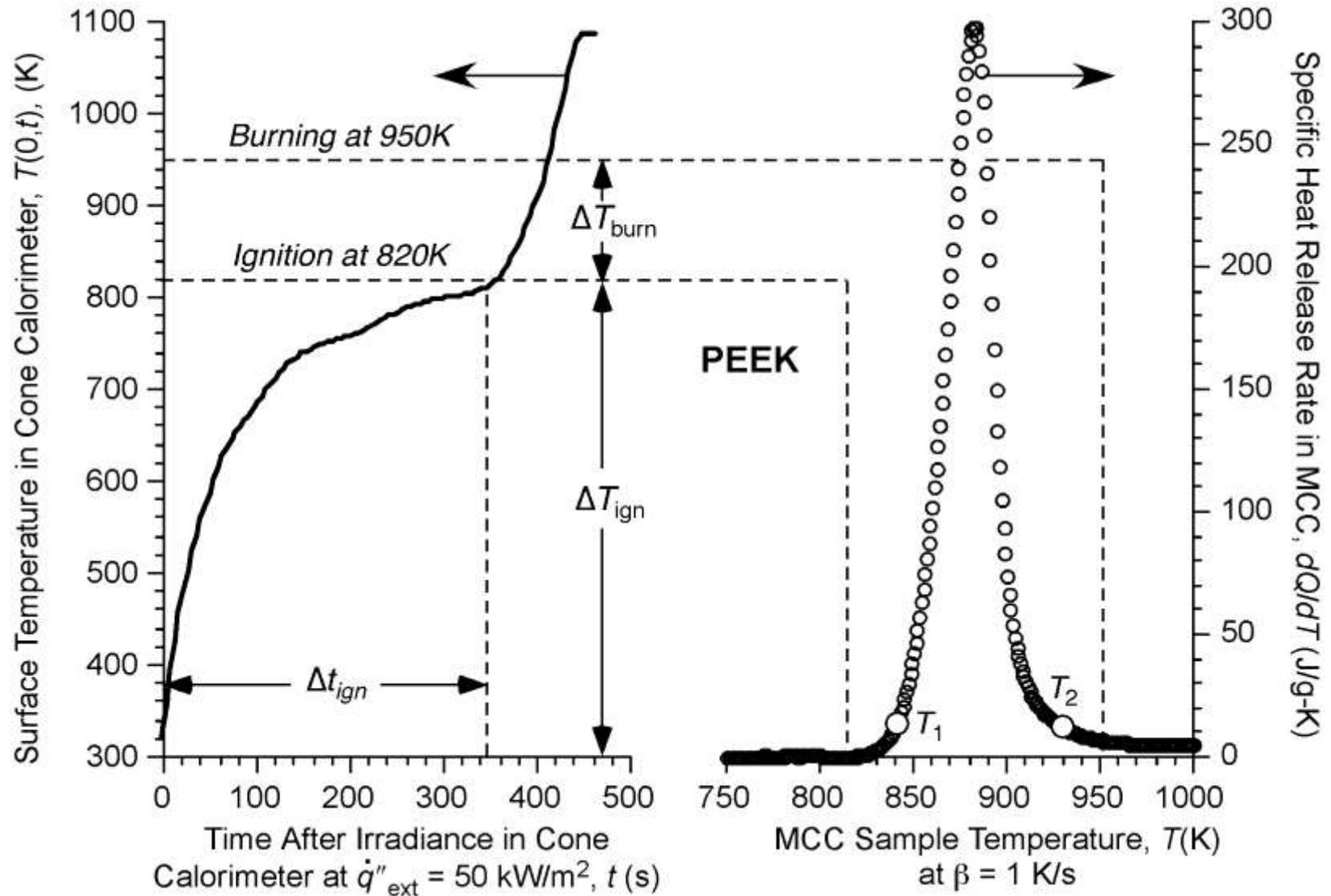
Ignition Temperatures: MCC \approx Cone



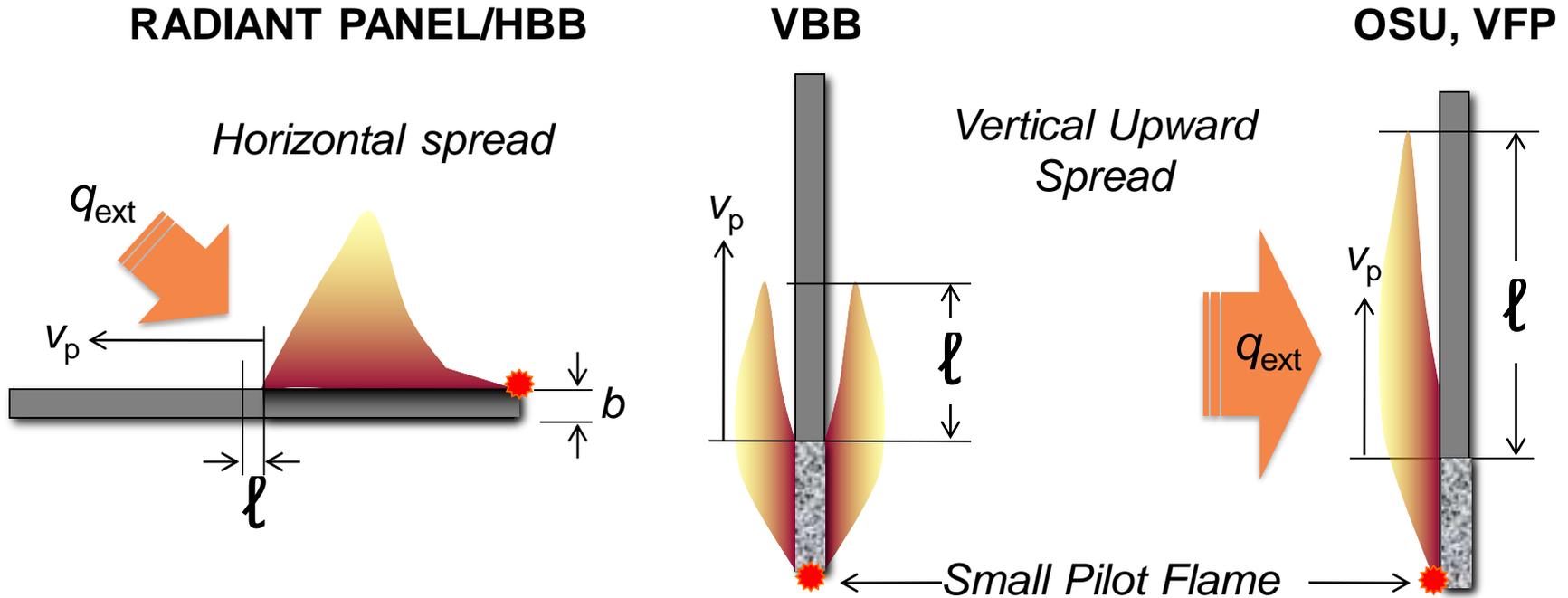
Fire Temperatures in Cone Calorimeter and MCC (Measured)



Fire Temperatures in Cone Calorimeter and MCC (Measured)



Fire Growth at Bench-Scale is 2-D



$$\begin{aligned}
 \text{Flame Spread Velocity, } v_y &= \frac{\text{Heated Length}}{\text{Time to Ignition}} = \frac{\ell}{t_{ign}} \propto \frac{\dot{q}''}{\Delta T_{ign}} \\
 &= \\
 \text{Burning Velocity, } v_x &\propto \frac{\dot{q}''}{\Delta T_{burn}}
 \end{aligned}$$

Fire Growth Capacity (FGC)

$$FGC = Q_{\infty} \left(\frac{1}{T_b - T_p} + \frac{1}{T_p - T_0} \right)$$

$$= \left(\frac{Q_{\infty}}{\Delta T_{burn}} \right) \left(\frac{H_g / c_p}{\Delta T_{ign}} \right)$$

$$\left(\text{Burning Rate} \right) \left(\text{Flame Spread Rate} \right)$$

Symbol	TA Property	Fire Parameter
Q_{∞}	Heat of Complete Combustion	Effective Heat of Combustion
T_p	Incipient Pyrolysis Temperature	Ignition Temperature
T_b	Final Pyrolysis Temperature	Burning Temperature
T_0	Ambient Temperature	
c_p	Heat Capacity of Solid	
H_g	Heat of Gasification of Solid	

Accurate FGC by Correcting for Baseline Drift in the MCC

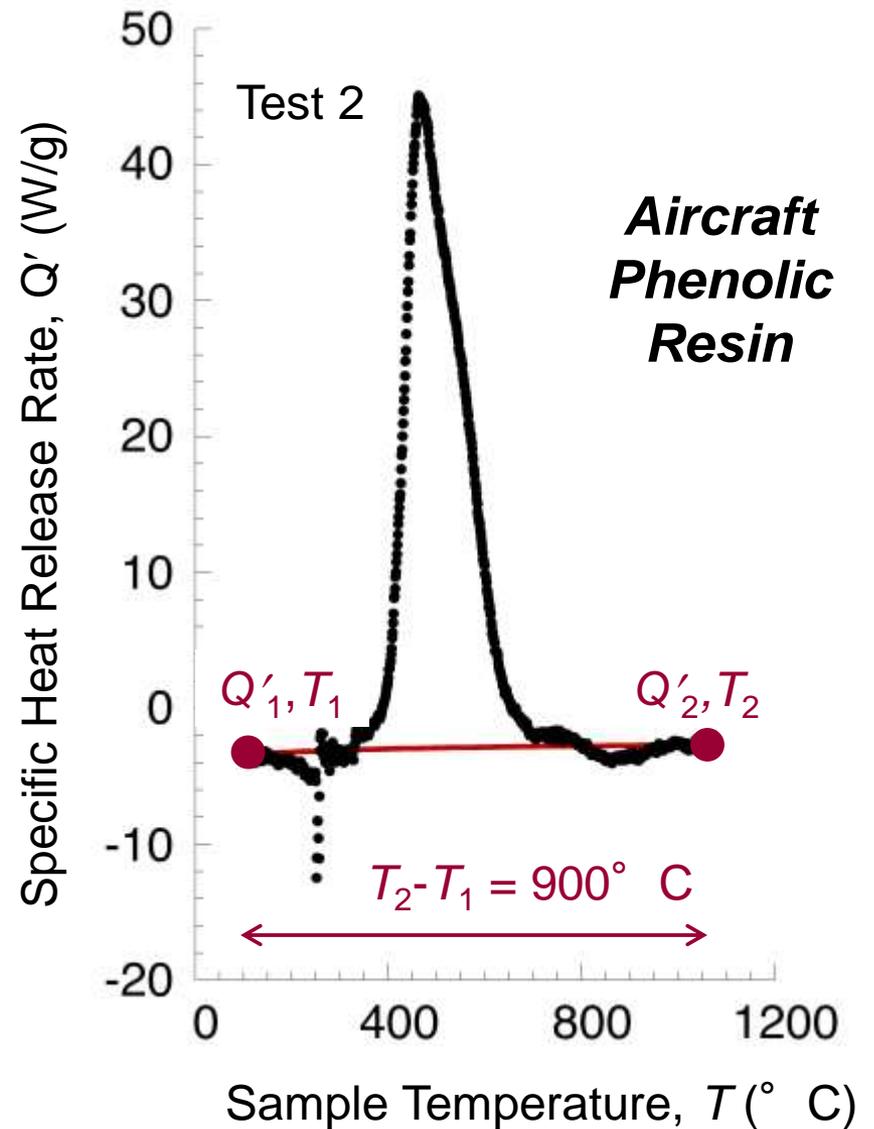
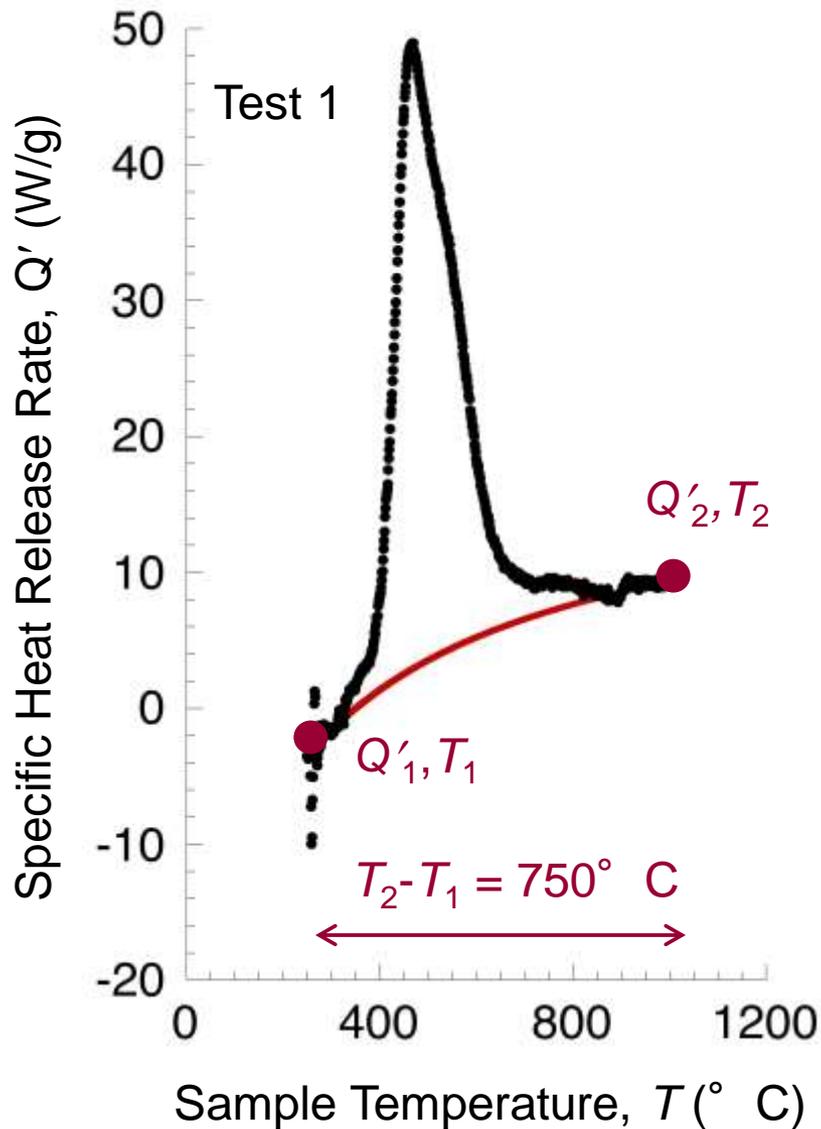
Thermal expansion of the purge gas in the pyrolyzer during the test has the effect of:

- Increasing the terminal flow rate (Methods A and B).
- Diluting O₂ at the sensor (Method A).

This change in the *zero-point* value of the specific heat release rate, Q'_0 is described by the ideal gas law, with T the sample/pyrolyzer temperature at $Q'(t)$

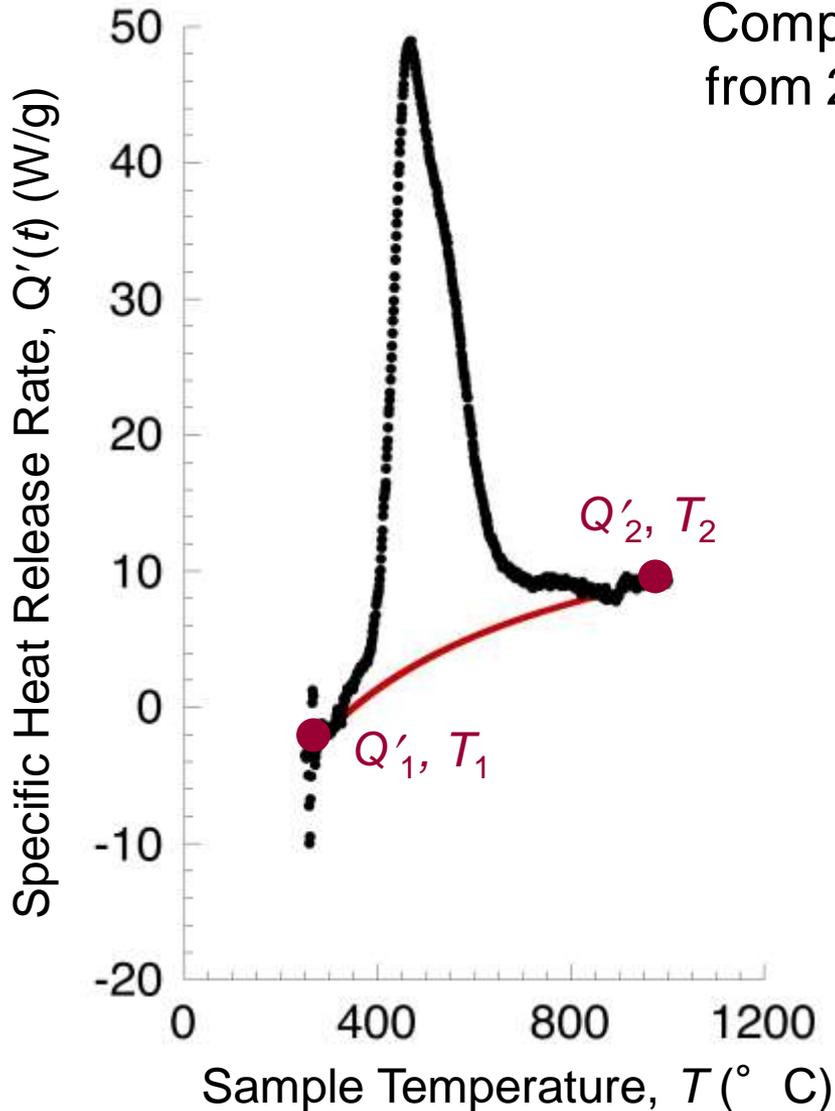
$$\text{MCC Baseline} = Q'_0 = \frac{C_1}{T} - C_2$$

At High Sensitivity Baseline Drift is Mainly Due To O₂ Sensor Fluctuations



Correcting MCC Data for Baseline Drift

Compute Baseline Coefficients, C_1 , C_2
from 2 data points, (Q'_1, T_1) and (Q'_2, T_2)



$$C_1 = \frac{Q'_1 - Q'_2}{(T_2 - T_1) / T_1 T_2}$$

$$C_2 = \frac{Q'_1 - Q'_2}{(T_2 - T_1) / T_2} - Q'_1$$

The specific heat release rate at
sample temperature T is:

$$Q'(T) = Q'(t) - Q'_0 = Q'(t) - \frac{C_1}{T} + C_2$$

Baseline Correction to D7309-19 is Balloted Item in ASTM D20.30 Subcommittee on Thermal Properties of Plastics

12. Calculation or Interpretation of Results

12.1. Calculate the specific heat release rate, $Q(t)$ (W/g) at time t during the heating program as per (15):

$$Q(t) = \frac{F}{m} \left(F_1 X_{1,1} + F_2 X_{1,2} \left[1 - \frac{1}{2} (X_{2,1} - X_{2,2}) \right] \right) \quad (15)$$

If H_2O and CO_2 are both removed from the combustion stream prior to measuring F and $X_{1,i}$, see Annex A2.

12.2. For Method A only:

12.2.1 The zero-point value (baseline) of $Q(t)$ changes during the normal operation of the instrument due to systematic thermal expansion of N_2 into the combustor during ~~sample~~ heating, and random O_2 sensor fluctuations. Systematic curvature of the $Q(t)$ baseline shall be corrected by selecting a value Q_1 at sample temperature T_1 immediately preceding combustion, and a value Q_2 at sample temperature T_2 immediately following combustion. The baseline will pass through the

8



points (Q_1, T_1) and (Q_2, T_2) in the $Q(t)$ history. Two baseline coefficients are calculated from these points at $20K$,

$$C_1 = (Q_1 - Q_2) / (T_2 - T_1) / T_1 T_2$$

$$C_2 = (Q_1 - Q_2) / (T_2 - T_1) - Q_1$$

The coefficients C_1 and C_2 are then used to calculate the baseline-corrected specific heat release rate $Q_c(T)$ at sample temperature T from the specific heat release rate at time t ,

$$Q_c(T) = Q(t) + C_1/T + C_2$$

Figure 3 illustrates the baseline correction process and result for a 6mg sample of phenolic resin.

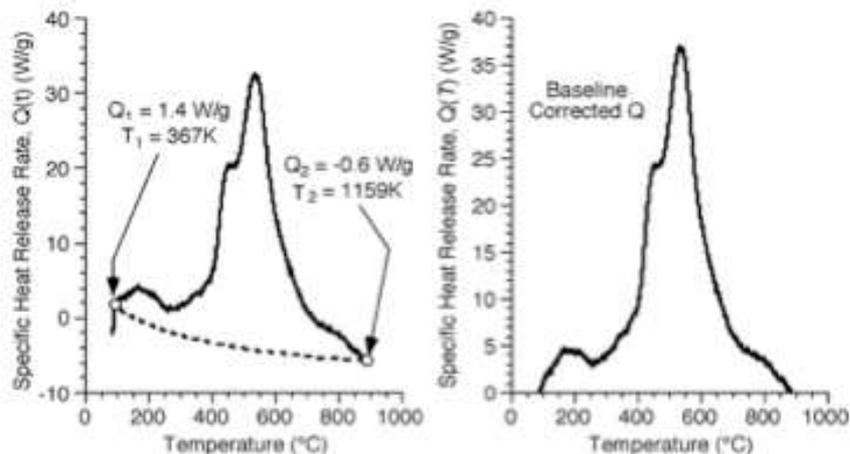
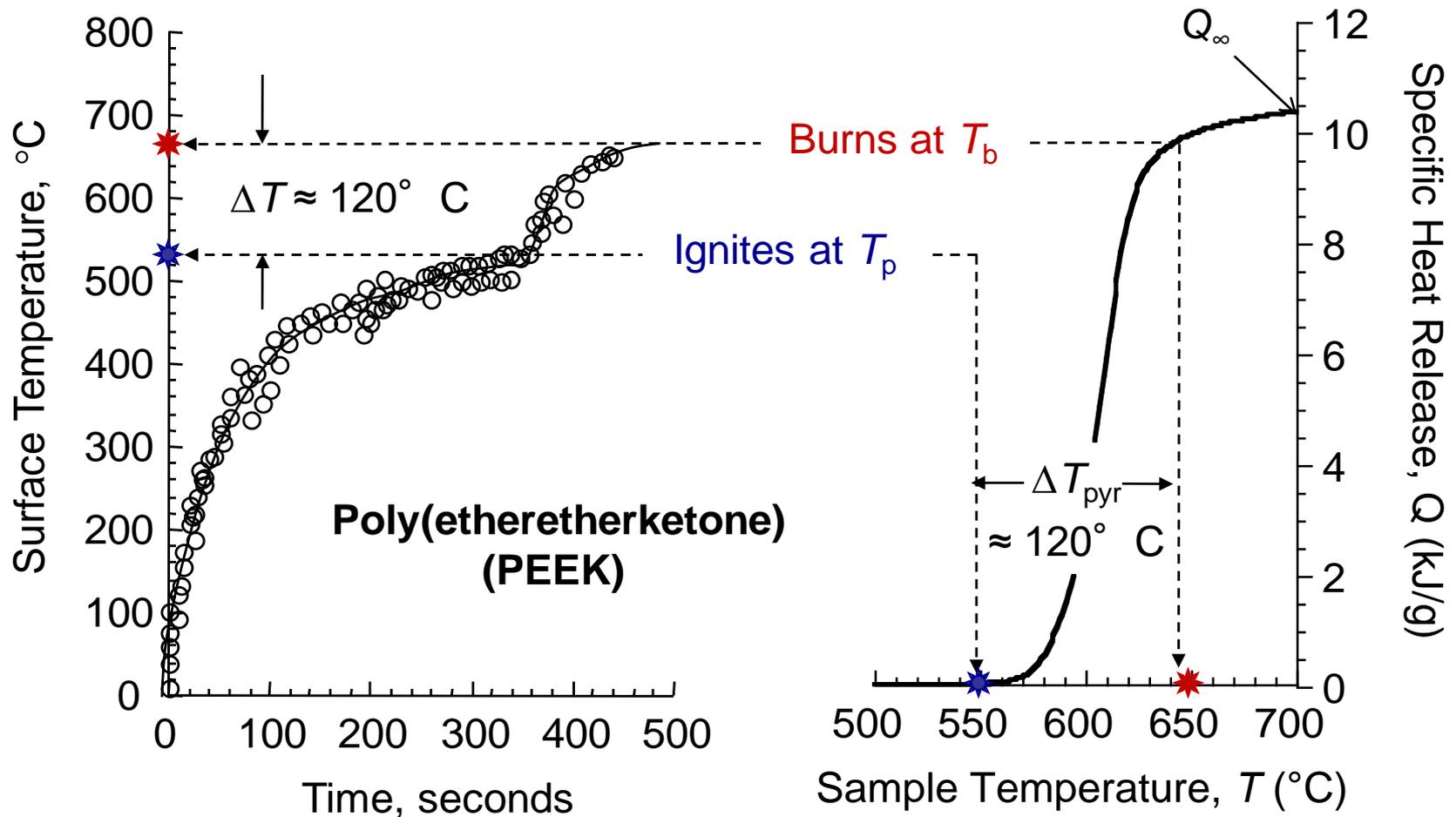


Fig 3 Baseline correction process and result for a 6mg sample of phenolic resin

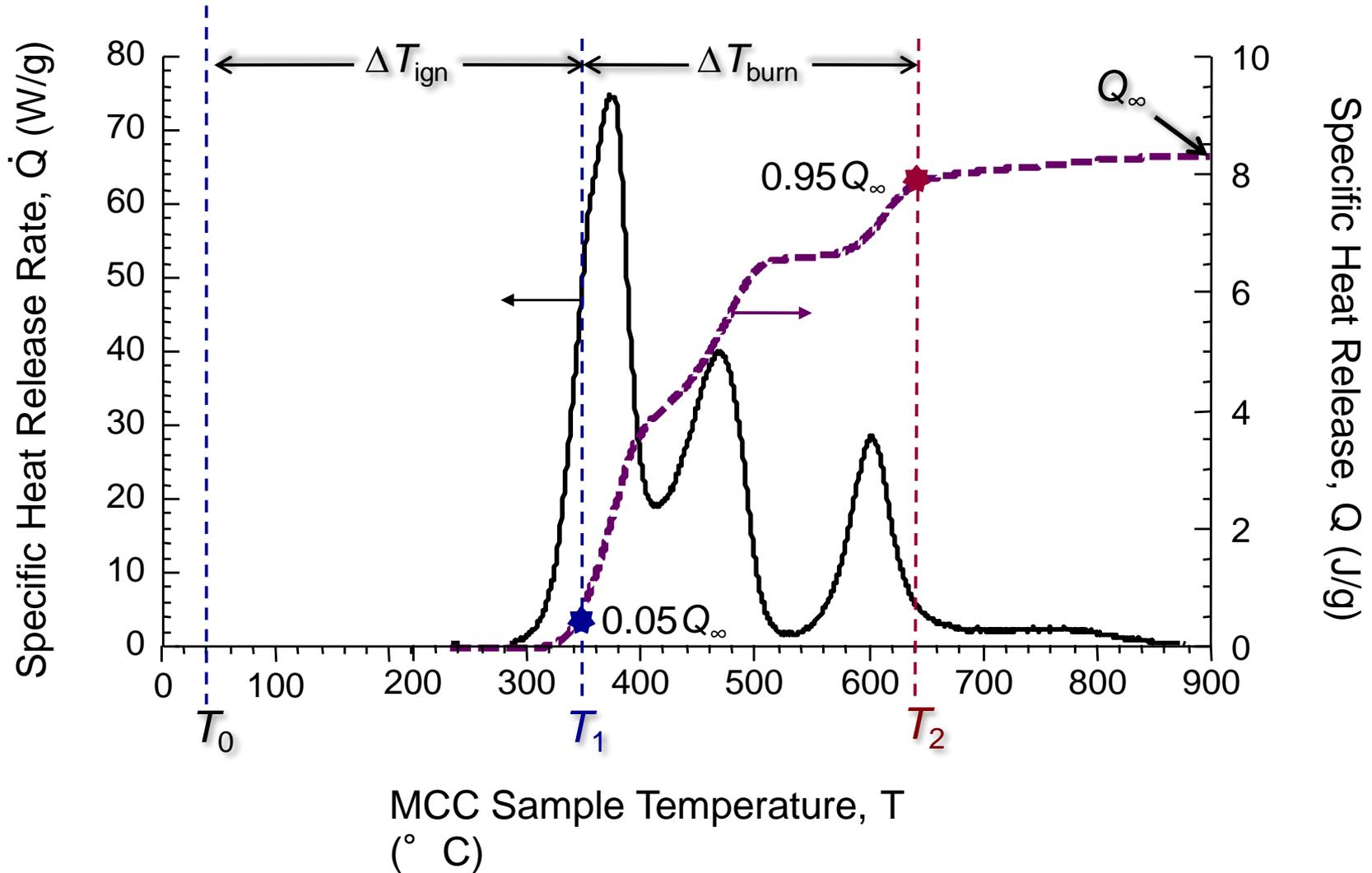
Ignition and Burning Temperatures

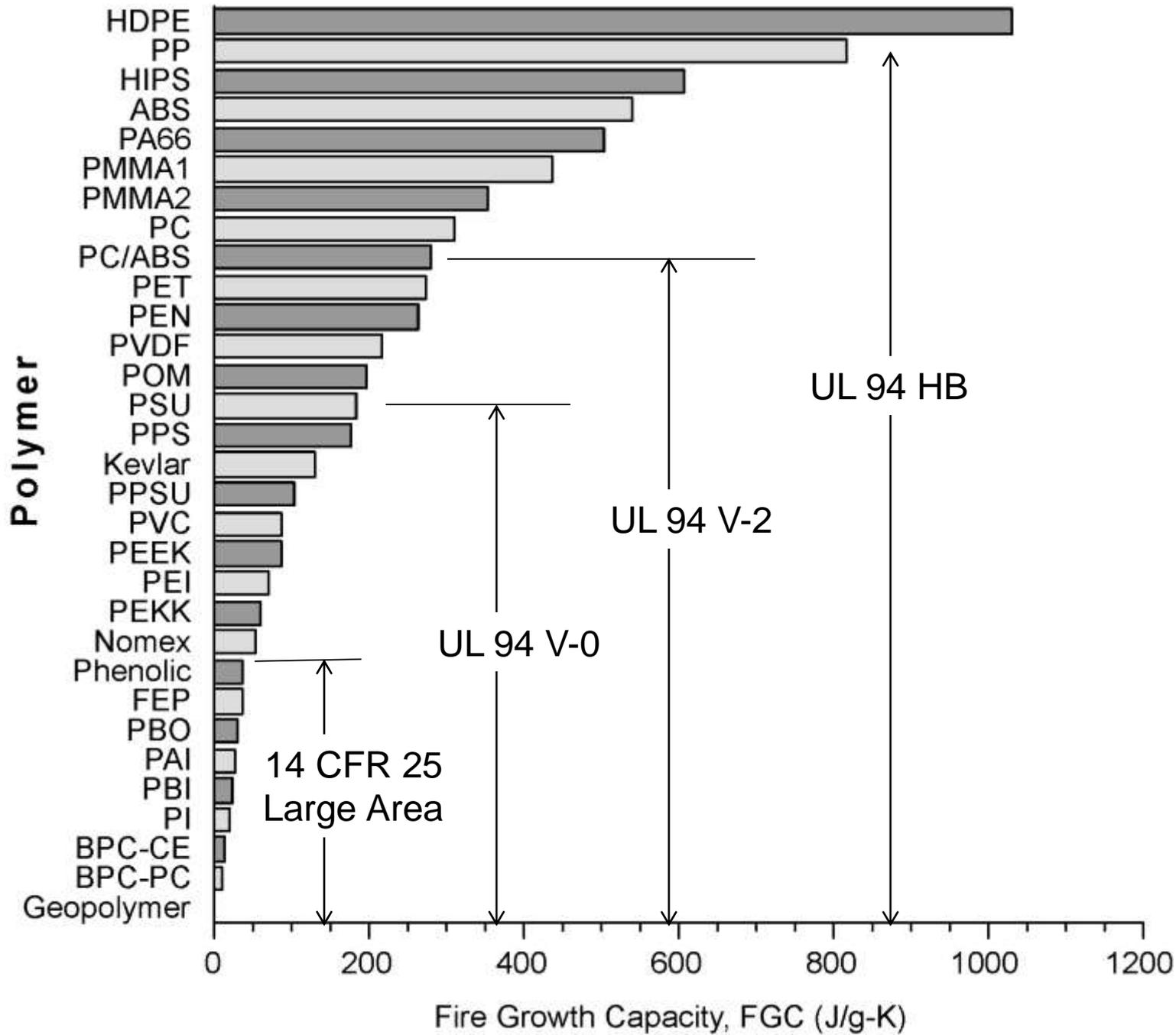
(From MCC Integral)



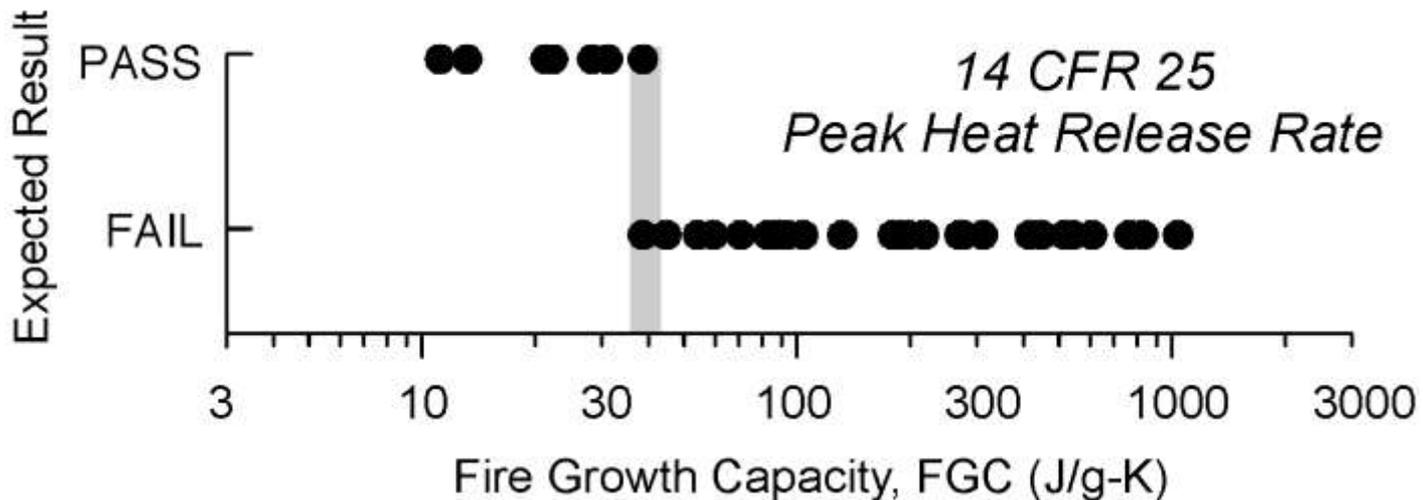
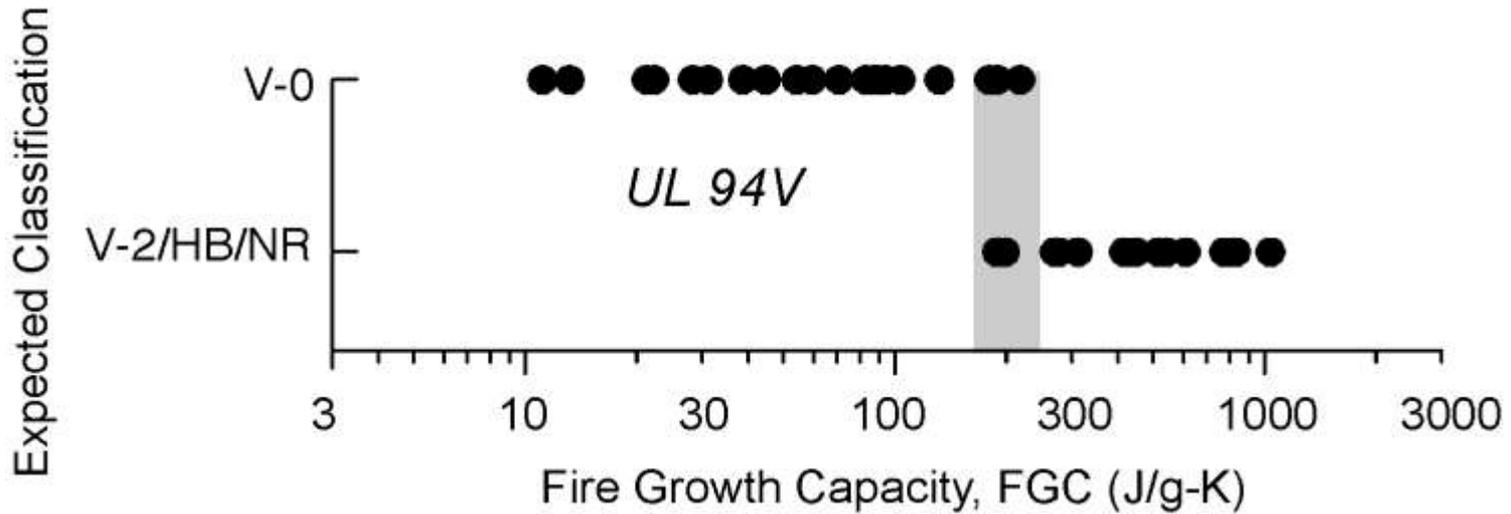
Q_∞ , T_p and T_b are TA properties in Fire Growth Capacity/FGC.

MCC Integral Method for FGC



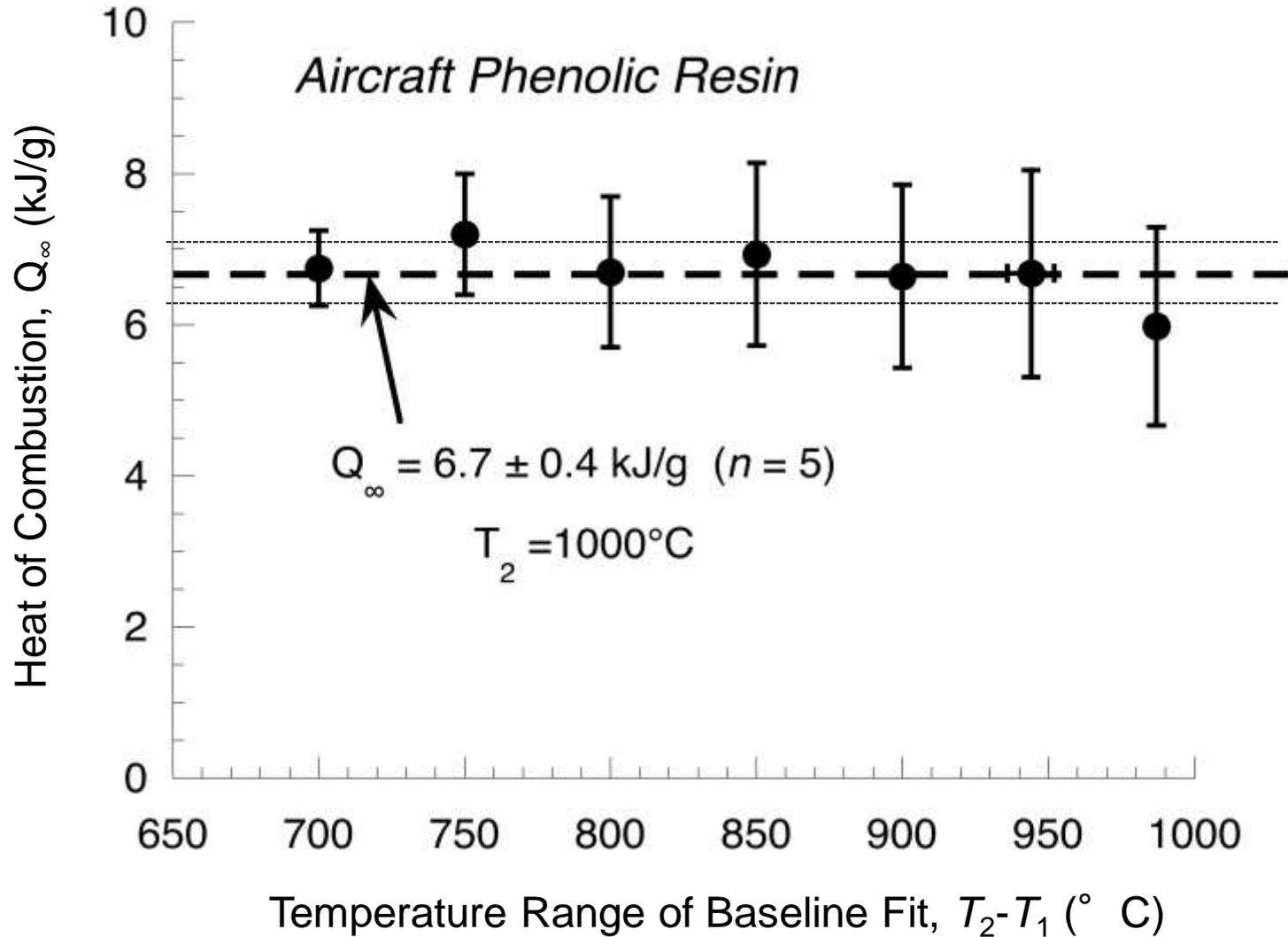


FGC Discriminates Levels of Fire Performance

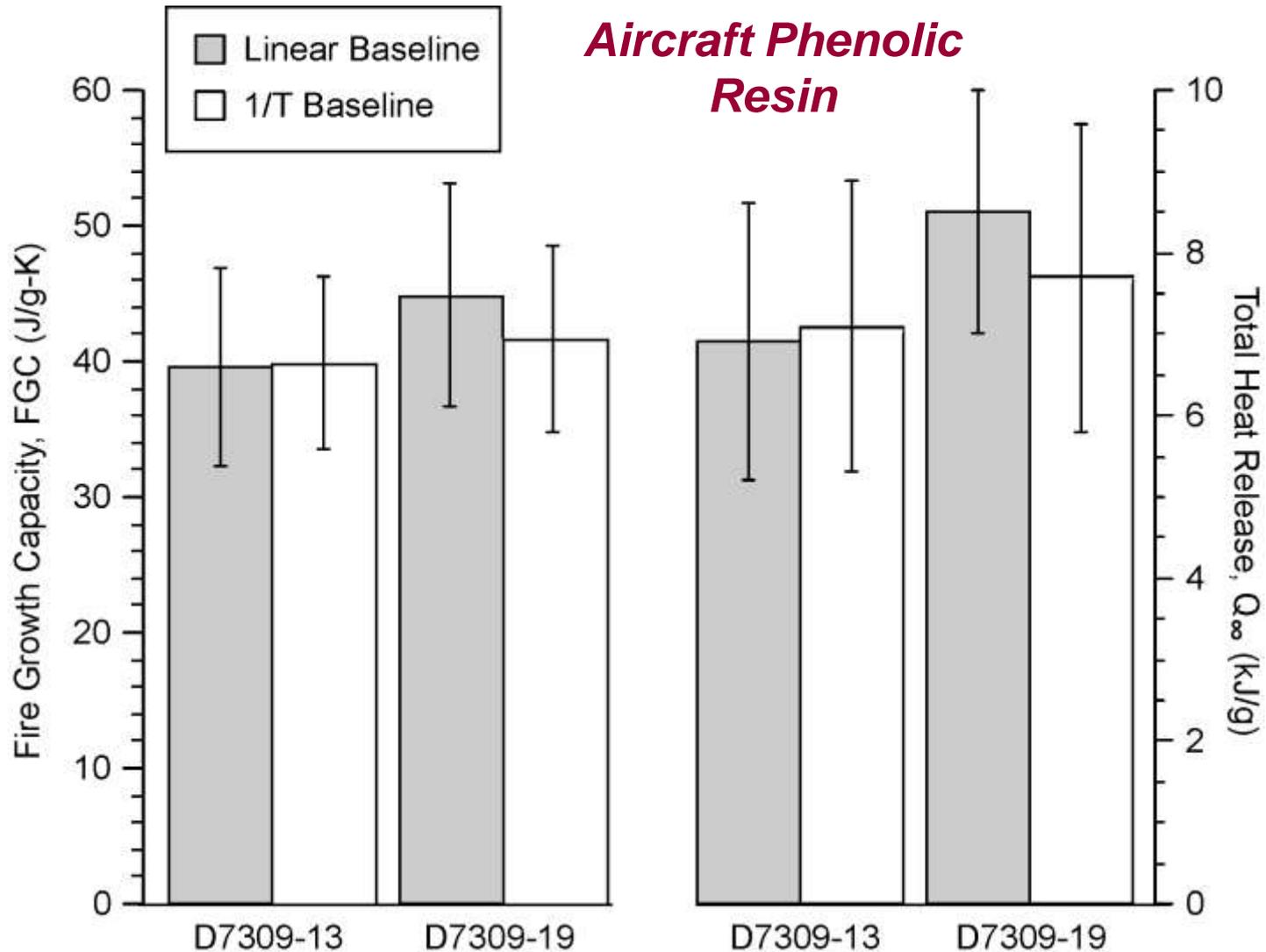


FGC is Independent of Choice of Baseline

$$\Delta T = T_2 - T_1$$



FGC is Independent of D7309 Version (2013/2019) And Baseline Method (Linear/ T^{-1})



Conclusions

Ignition temperature (T_{ign}) and burning temperature (T_{burn}) of components identified in MCC.

MCC data corrected for baseline drift to obtain accurate T_{ign} , T_{burn} and total heat Q_{∞} of combustion for similarity determination.

Fire growth capacity (FGC) combines flame spread and burning rate in a single parameter and is useful for comparing flammability of polymers at micro (10^{-6} kg) scale.

Microscale (FGC) criteria for equivalent bench (kg) scale flammability of certified and substitute components has been demonstrated (Safronava).