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HAZARDOUS PRODUCT PRODUCED BY ELECTRICAL INSULATION

IN BRIEF ... Generation of HC ℓ occurs when PVC cable insulation is heated, externally or by internal heating induced by electric overload. The HC ℓ vapor may be present in the atmosphere, at hazardous levels, prior to the generation of black smoke--so that fire-fighting personnel may not be readily cognizant of the danger.

A recent study at the NBS Building Research Division (U.S. Department of Commerce), has shown that when polyvinyl chloride (PVC) wire insulation is heated it goes through two stages of decomposition. In the initial stage, when the level of heat input is relatively low, a white HC ℓ mist is generated. In the latter phase, when the material is exposed to higher levels of heat input or an open flame, a dark sooty smoke is released.

The release of the HCl prior to the development of dense smoke presents a potentially hazardous situation, especially to fire-fighting personnel. The HCl in this state is colorless, but forms a white mist resembling steam by combining with moisture in the air. Although innocuous in appearance, the HCL acts as a primary irritant and corrosive agent to the respiratory system.

The decomposition phases of PVC are a function of the temperature of the material. Gaseous hydrogen chloride is released from PVC when the temperature reaches about 450 $^{\circ}$ F. The rate of release depends upon the temperature and density of the material. The gas, on exposure to moisture in the air, forms white fumes of acid vapor by absorption of moisture. The vapor is very reactive and can be easily adsorbed on contact with a surface. If the temperature is maintained below 650 $^{\circ}$ F, dark smoke or soot will not form. Above 650 $^{\circ}$ F, carbonaceous degradation products in the form of black smoke are present.

In the NBS study a specimen of an electric cable covered by black plastic insulation, composed mainly of PVC with some amounts of fillers and plasticizers, was tested. The insulation was cut and flattened to fit in the standard holder in a smoke density chamber. $\frac{1}{}$ The specimens were irradiated with 2.5 watts/cm² heat energy for non-flaming exposure tests.

Results were obtained in terms of optical density and quantity of HCl concentration (fig. 1). In the 18 cu ft chamber, the combustion products from a 6.5 sq in sample reached a HCl concentration of above 1 600 ppm. From 1 000-2 000 ppm of HCl is considered to be dangerous

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on short exposure, 2^{-1} whereas 5 000 to 10 000 ppm of carbon monoxide are needed for comparable effect. Firefighting personnel should thus be alert to the possible dangers of electrical insulation decomposition in a fire situation.

<u>1</u>/ Gross, D., Loftus, J. J., Robertson, A. F., Method for Measuring Smoke from Burning Materials, ASTM STP No. 422.

2/ Henderson and Haggard, Noxious Gases, Reinhold Publishing Co.



HCL CONCENTRATION, P.P.M.

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CAPTION FOR ILLUSTRATION:

Figure 1. HCL concentration and optical density of PVC

electrical insulation decomposition products are shown in these plots. The data were obtained in an NBS study that showed hazardous concentrations of HCL can be released when polyvinyl chloride wire insulation is heated.

Attached

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