Polysilphenylenes

U. Lauter, S.W. Kantor, K. Schmidt-Rohr and W.J.MacKnight Polymer Science and Engineering Department, University of Massachusetts Amherst, MA 01003

Research Objective: The motivation for this work is to develop novel fire retardant elastomers based on vinyl substituted silphenylene siloxane copolymers.

Approach: The motivation to investigate flame retardant elastomers by substituting a silphenylene siloxane polymer backbone with a high number of vinyl groups is the following: Vinyl substituted siloxanes can be cured by addition of a multifunctional Si-H compound under catalysis with a variety of platinum complexes. The hydrosilylation reaction of the double bonds with the Si-H cross-linker will then generate a network structure. Depending upon the ratio of cross-linker to vinyl groups, the cross-link density of such a system can be controlled over a broad range. Materials can thus be obtained with properties ranging from a slightly cross-linked elastomer to a highly cross-linked resin type structure. For elastomers, our concept is to transform a polymer with high vinyl content by hydrosilylation of a small amount of all vinyl groups present. The bulk of the unsaturated functionalities will then remain and be available to thermally cross-link at temperatures higher than the application temperature of the elastomer, i.e. in case of a fire. The resulting resin structure will then guarantee high char yields and, hence, good flame retardancy.

Accomplishment Description: The synthesis of a series of vinyl substituted silphenylene siloxane polymers of high molecular weight is described. 29Si NMR spectroscopy indicates that the polymers possess an exactly alternating microstructure of the silphenylene and siloxane moieties. The materials are completely amorphous with glass transition temperatures between -60° C and -86° C. All polymers degrade thermally only above 500°C under both oxidative or inert atmosphere. The char yields at 900°C increase with increasing vinyl content from 30 percent up to almost 70 percent. Cross-linking of a small fraction of vinyl groups by hydrosilylation gives materials with elastomeric properties. Dynamic mechanical experiments indicate that the bulk of vinyl groups cross-links thermally at temperatures above 230°C. Therefore, the temperature window for possible applications of these materials as fire-safe elastomers extends from about -86° C to approximately 230°C.

Significance: A series of strictly alternating vinyl substituted (i.e. cross-linkable) silphenylene siloxane copolymers with systematically varying vinyl content has been synthesized. The polymers are completely amorphous gums with glass transition temperatures ranging from -86° C to -59° C as a function of vinyl content and volume fraction of highly mobile siloxane units. The materials degrade thermally above 500°C. As the vinyl content increases, both the onset temperature of major degradation and the char yield at 900°C shift to higher values.

Expected Results: One of the polymers has an extremely high degradation temperature of over 600° C and a char yield of almost 70 percent in air, which makes it a promising candidate for potential applications as a flame retardant elastomer. The reason for the

remarkable thermal stability is ascribed to a cross-linking reaction of the vinyl functionalities, which occurs slightly below the onset of siloxane redistribution reactions around 300° C.

References: Dvornic, P.R., Lenz, R.W., High Temperature Siloxane Elastomers, (Huethig & Wepf Verlag, New York, 1990).

Folland, P.D., Matisons, J.G., "Siloxane Elastomers" in J.C. Salamone (Ed.), Polymeric Materials Encyclopedia, (CRC Press, 1996).

Lauter, U., Kantor, S.W., Schmidt-Rohr, K., and MacKnight, W.J., "Vinyl Substituted Silphenylene Siloxane Copolymers -- Novel High Temperature Elastomers", Macromolecules (in press).

Point of Contact: Dr. Richard E. Lyon, AAR-422, FAA William J. Hughes Technical Center, Atlantic City International Airport, NJ 08405, (609) 485-6076, FAX: (609) 485-6909, email: rlyon@admin.tc.faa.gov.