

ENVIRONMENTALLY-BENIGN FLAME RETARDANT COATINGS FOR POLYMERS

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(Presented by Natalie Vest)

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*10th Triennial International Aircraft Fire and Cabin Safety
Research Conference – Atlantic City, NJ – 18 October 2022*



Polymer NanoComposites (PNC) Lab (<http://nanocomposites.tamu.edu>)

Scientific Pillars:

❖ **Polyelectrolyte complexation / assembly**

❖ **Polymer-nanoparticle interactions**

❖ **Water-based processing**

❖ **Renewable chemistry**

Nature Rev. Mater. **2020**

ACS Mater. Lett. **2020**

ACS AMI **2018**

J. Mater. Sci. **2017**

Adv. Mater. Interf. **2015**

Advanced Materials **2011**

ACS Nano **2009**

Adv. Mater. Interf. **2019**

Macro. Rapid Comm. **2017**

Green Materials **2016**

Macromolecules **2015**

Langmuir **2015**

Macro. Rapid Comm. **2015**

ACS Macro Lett. **2014**

Adv. Electronic Mater. **2019**

Advanced Materials **2018**

Nano Energy **2016**

Adv. Energy Mater. **2016**

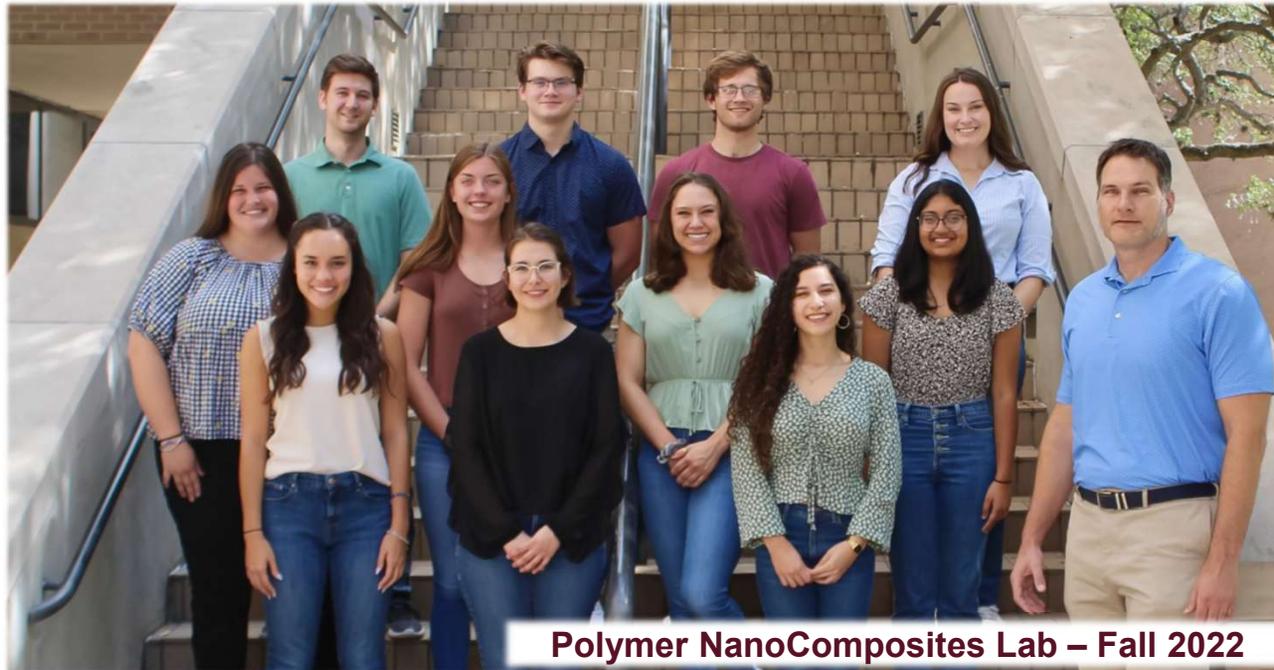
Advanced Materials **2015**

ACS Nano **2010**

Nano Letters **2008**



Acknowledgements



Polymer NanoComposites Lab – Fall 2022

Collaborators:

- Sandra Bischof (Univ. Zagreb)
- Serge Bourbigot (Univ. Lille)
- Federico Carosio (Pol. Torino)
- Steve Eichhorn (Bristol)
- Sabysachi Gaan (Empa)
- Igor Jordanov (N. Macedonia)
- Alex Morgan (UDRI)
- Maja Radetic (Serbia)
- Mohammad Naraghi (TAMU)
- Oren Regev (Ben Gurion U.)
- Patrick Shamberger (TAMU)
- Lars Wagberg (KTH)
- Xin Wang (USTC)
- Anthony Yuen (UNSW)

☩ Psalm 19:1-6

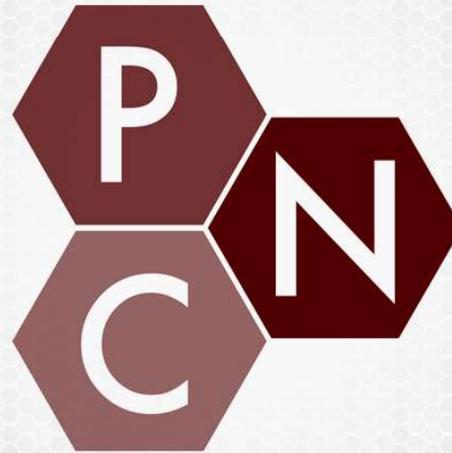


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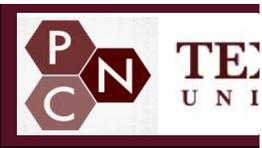
Polymer NanoComposites (PNC) Lab (<http://nanocomposites.tamu.edu>)



- ✘ Overview of polyelectrolyte complexes (PEC) in water
- ✘ Flame retardant treatments from “complex” coatings
- ✘ Heat shielding from layer-by-layer nanobrick wall coatings



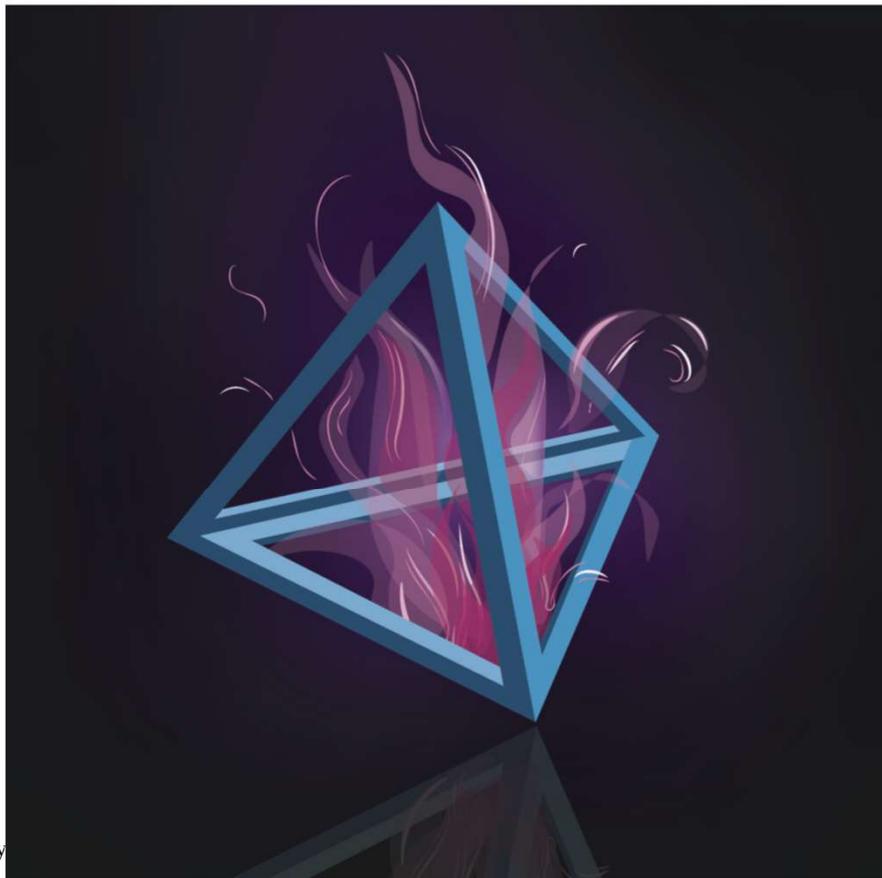
Polymer NanoComposites (PNC) Lab (<http://nanocomposites.tamu.edu>)



April 2020 volume 5 no. 4
www.nature.com/natrevmats

Review of flame retardant surface treatments

nature reviews materials



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Review of flame retardant nanocoatings
developed using layer-by-layer assembly of
polyelectrolytes

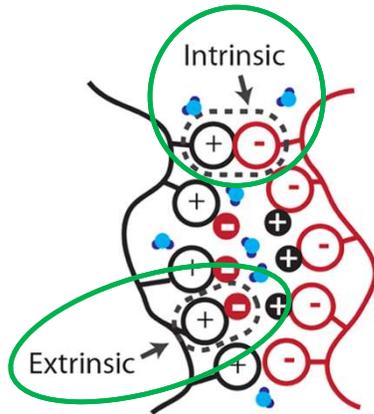
Fire Retardant Facts

**M. Holder, Ryan J. Smith & Jaime
García**

Journal of Materials
Journal of Materials

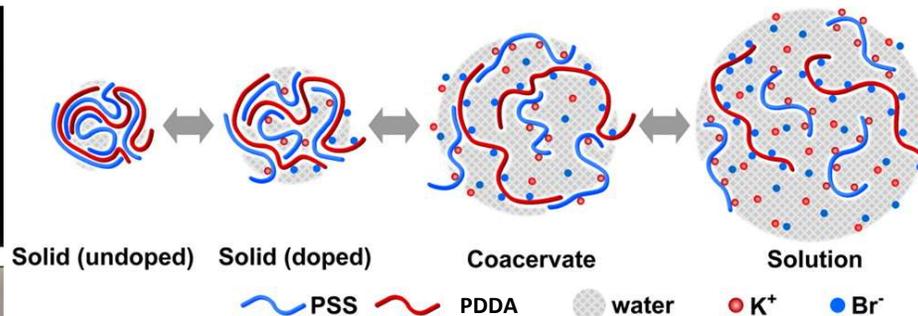
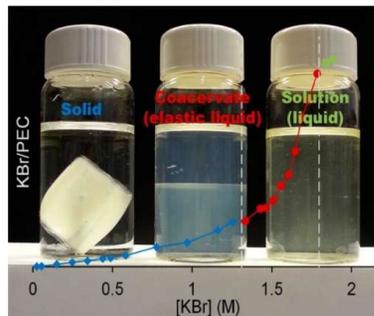
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12:017-1390-1





Coulombic interactions cause polyelectrolyte complexation (PEC)

- Entropic driving force through expulsion of small counter ions and water
- PEC form along a spectrum from insoluble complex to soluble solution



Zhang, Y., et al. *ACS Cent. Sci.* **2018**, *4*, 638.

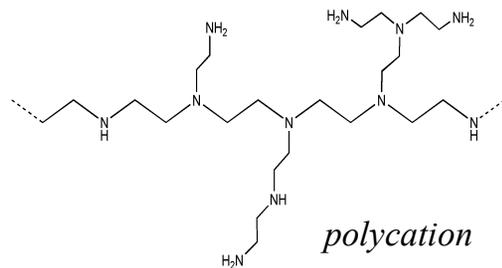
Wang, Q., et al. *Macromolecules*, **2014**, *47*, 3108.

Chiang, H.-C.; Grunlan, J. C.; et al. *Macromol. Rapid Comm.* **2021**, *42*, 2000540.

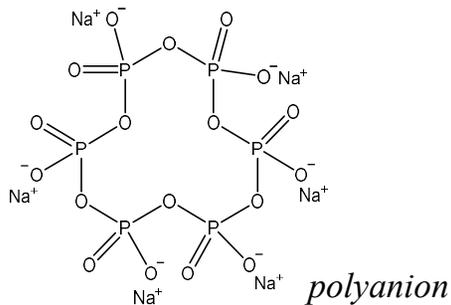
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Polyethylenimine (PEI)



Sodium hexametaphosphate (PSP)



- Chemistry helps to form insulating char
- pH affects PEI degree of protonation
- Polyelectrolytes flocculate at $\text{pH} \leq 8$, but mutually suspended above $\text{pH} 9$

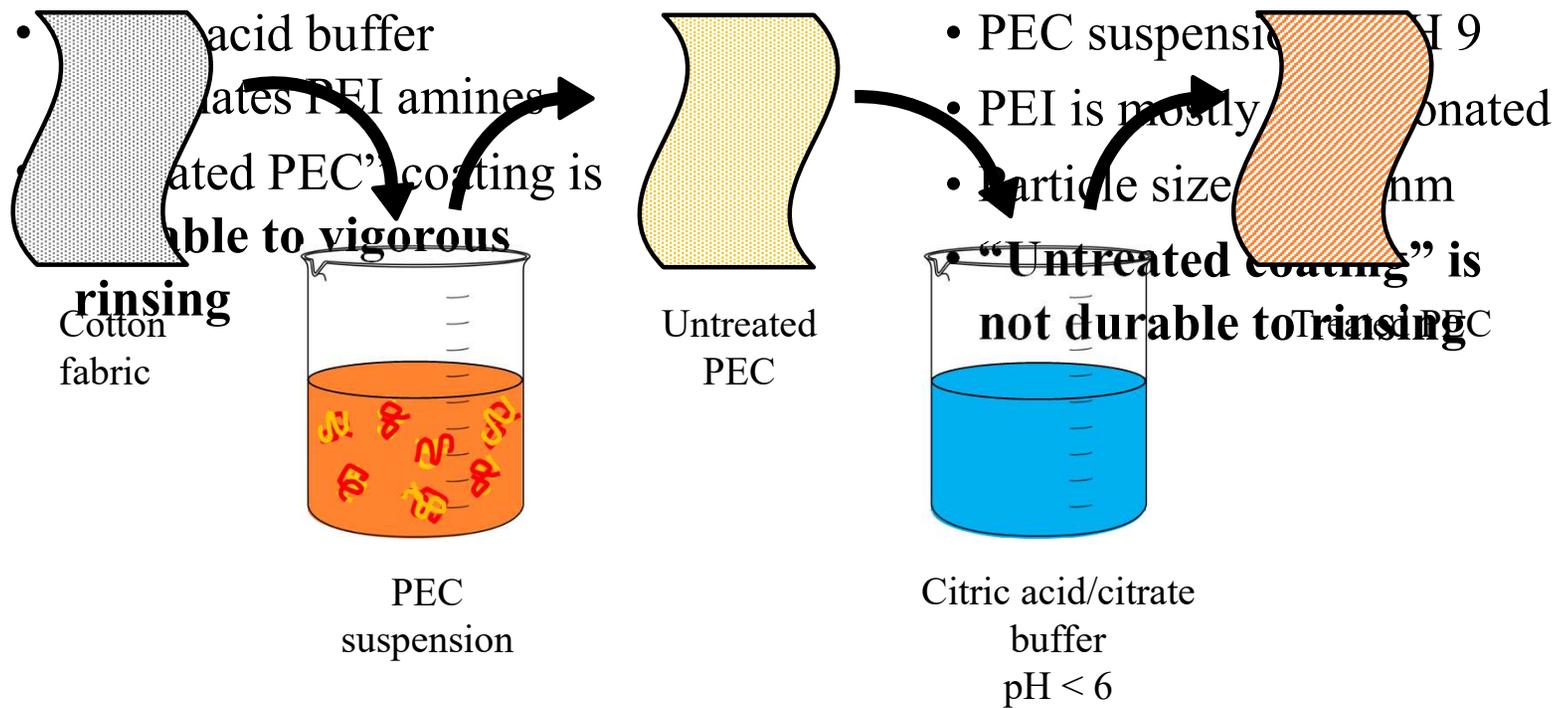
M. Haile, C. Fincher, S. Fomete, J. C. Grunlan, *Polym. Degrad. Stab.* **2015**, 114, 60–64.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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PEC Coating of Cotton Fabric



M. Haile, C. Fincher, S. Fomete, J. C. Grunlan, *Polym. Degrad. Stab.* **2015**, 114, 60–64.

J. C. Grunlan, “Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates,” U.S. Patent 9,840,629.

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Flame Testing of Cotton



Uncoated control



Untreated PEC coating



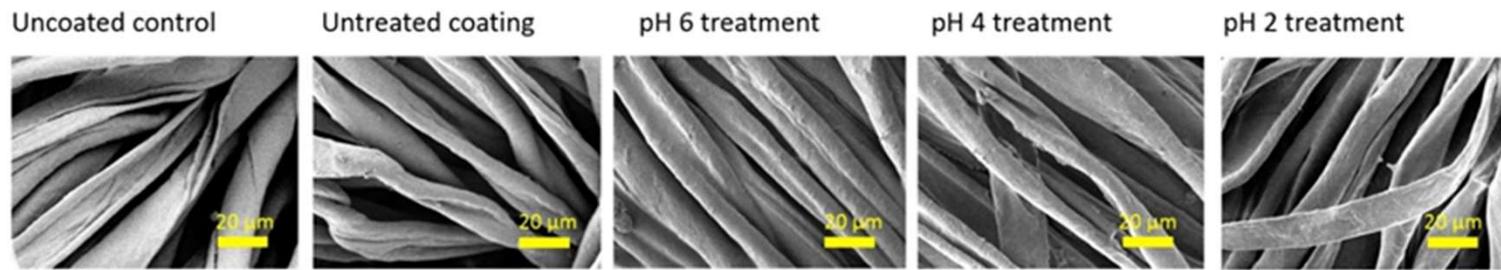
PEC coating treated by pH 2 buffer



M. Haile, C. Fincher, S. Fomete, J. C. Grunlan, *Polym. Degrad. Stab.* **2015**, 114, 60–64.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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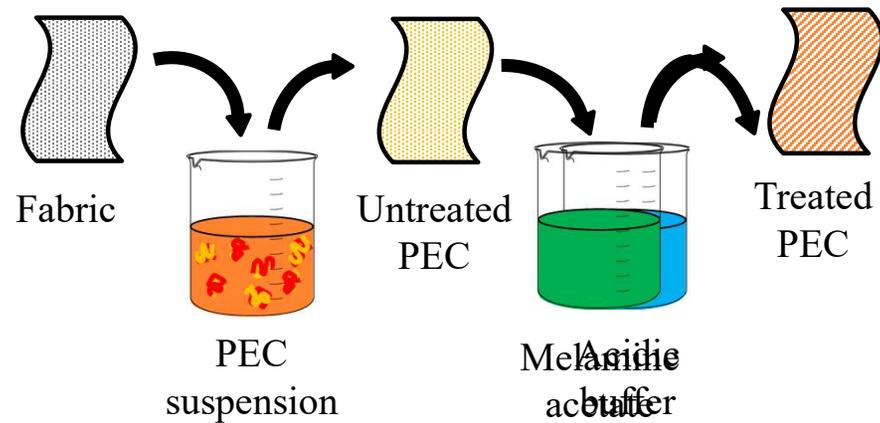


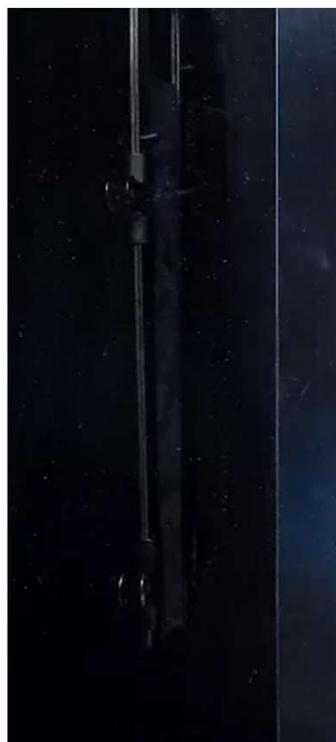
M. Haile, C. Fincher, S. Fomete, J. C. Grunlan, *Polym. Degrad. Stab.* **2015**, 114, 60–64.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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- Nylon-cotton (NYCO) fabric particularly challenging substrate
- Phosphate acts to catalyze the charring of cellulose
- Melamine polyphosphate can add further FR protection





No coating



19 wt% (PEC + MeI^{2%})

M. Leistner, M. Haile, S. Rohmer, A. Abu-Odeh, J. Grunlan, *Polym Degrad Stab*, **2016**, *122*, 1-7.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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control

PEC

PEC + MeI^{2%}

PEC + MeI^{5%}



PEC: 7 wt% PEI + 14 wt% APP

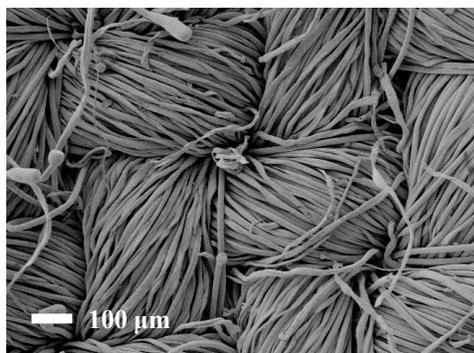
M. Leistner, M. Haile, S. Rohmer, A. Abu-Odeh, J. Grunlan, *Polym Degrad Stab*, **2016**, *122*, 1-7.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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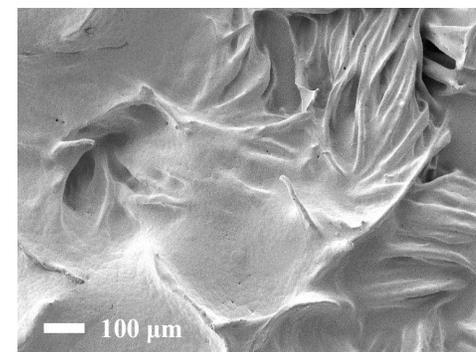
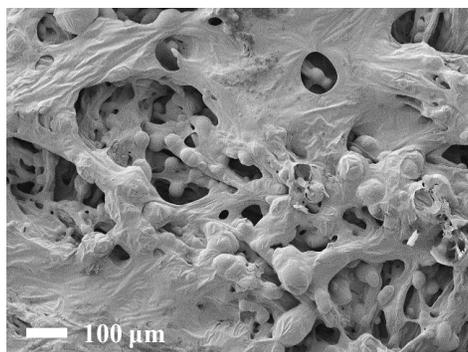
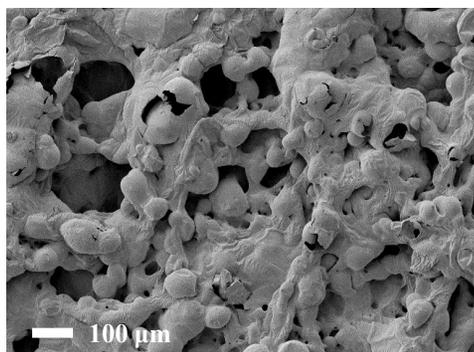
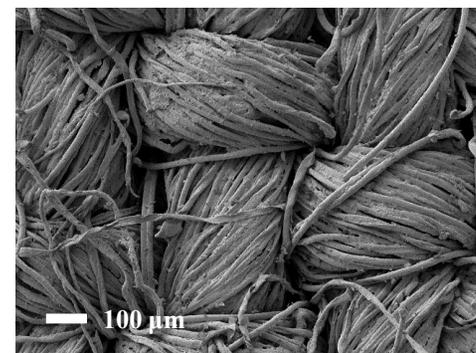
No coating



PEC



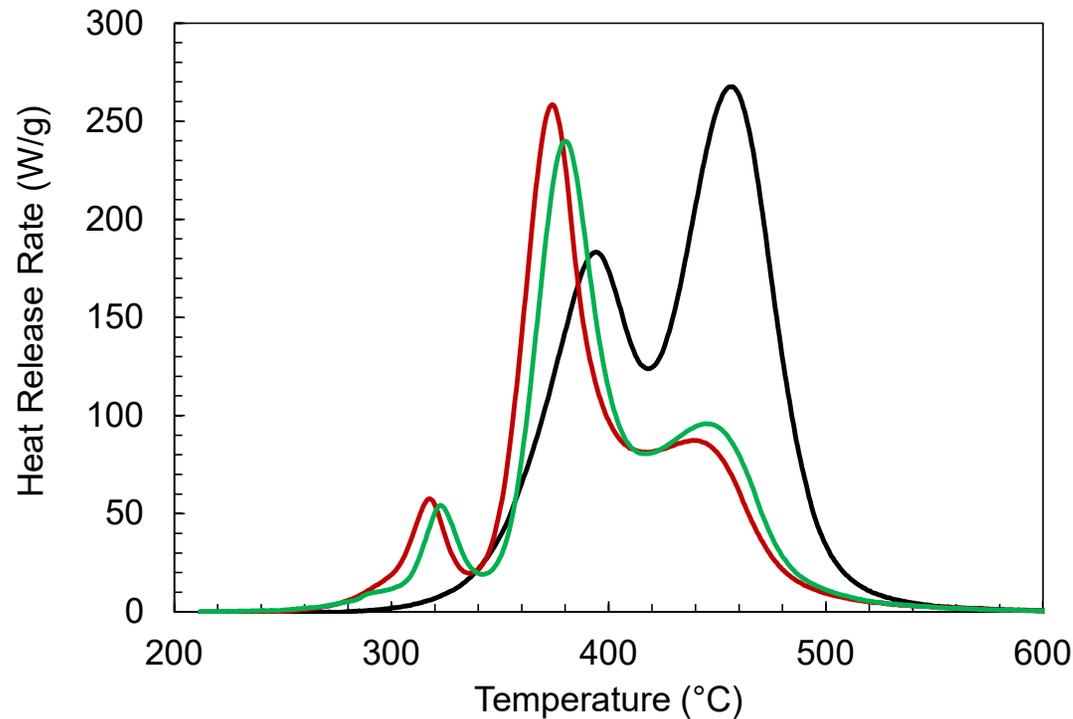
PEC + Mel^{2%}



Melamine addition creates strong, dense char that acts as heat shield and barrier to oxygen and volatiles.

M. Leistner, M. Haile, S. Rohmer, A. Abu-Odeh, J. Grunlan, *Polym Degrad Stab*, 2016, 122, 1-7.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," U.S. Patent 9,840,629.



No coating
2 peaks
THR: 19.1 kJ/g

PEC
3 peaks
THR: 13.6 kJ/g

PEC + Mel^{2%}
3 peaks
THR: 13.7 kJ/g

PCFC (aka MCC) not able to detect char density or cooling effects.

M. Leistner, M. Haile, S. Rohmer, A. Abu-Odeh, J. Grunlan, *Polym Degrad Stab*, **2016**, *122*, 1-7.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," U.S. Patent 9,840,629.

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	VFT	THR (by PCFC)	Energy balance* (260 – 500°C)
no coating	burned off	19.1 kJ/g	+ 340 J/g
PEC	burned off	13.6 kJ/g	+ 70 J/g
PEC + Mel ^{2%}	self-extinguishing	13.7 kJ/g	– 60 J/g

* measured by DSC in N₂ at a heating rate of 10 K/min

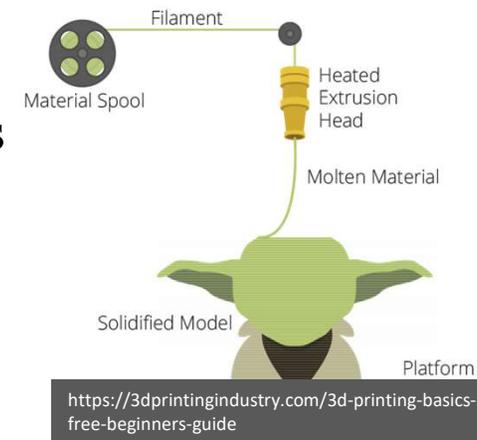
DSC reveals a change in the energy balance during pyrolysis that reveals melamine addition making a more endothermic situation.

M. Leistner, M. Haile, S. Rohmer, A. Abu-Odeh, J. Grunlan, *Polym Degrad Stab*, **2016**, *122*, 1-7.

J. C. Grunlan, "Aqueous Polyelectrolyte Complex as One Pot Nanocoating Solution to Impart Antiflammable Behavior to Various Substrates," **U.S. Patent 9,840,629**.

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- Fused Filament Fabrication
 - Filaments are flammable thermoplastics
 - Causes fires, limits part applications



is counterproductive for filaments
retardant not localized to surface of part

3D printer blamed for fire inside Cain Building on Texas A&M campus

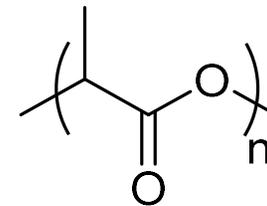
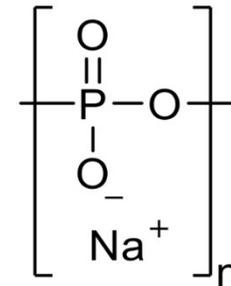
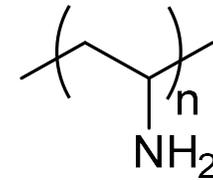
Firefighters say a 3D printer started a fire inside a classroom Wednesday afternoon inside the James J. Cain Building on the Texas A&M campus.

<https://kbtx.com>

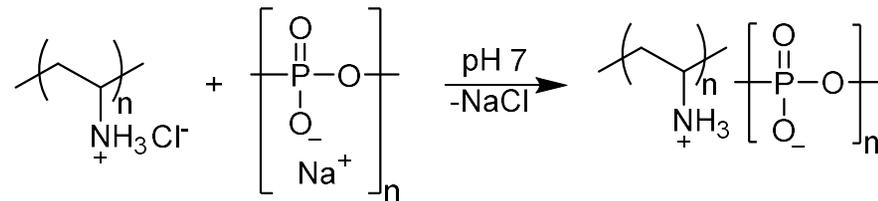
C. B. Sweeney, B. A. Lackey, M. J. Pospisil, T. C. Achee, V. K. Hicks, A. G. Moran, B. R. Teipel, M. A. Saed, M. J. Green, *Sci. Adv.* **2017**, *3*, e1700262.

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- Polyvinylamine (PVA)
 - BASF Lupamin 9095
 - Estimated $M \sim 205$ kDa
- Sodium hexametaphosphate (PSP)
 - $M_n \sim 3$ kDa (estimated)
- Polylactic acid (PLA)
 - 3D Solutech filament
 - Most common 3D printing filament

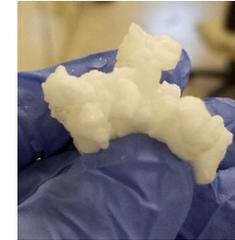


- Mix PVA & PSP
 - Separate solutions each pH 7, 0.25 M



- Dried overnight at 120 °C

- Resultant PEC can be extruded
 - Plasticize with DI water, extrude at 90 °C
- Intrinsically flame retardant

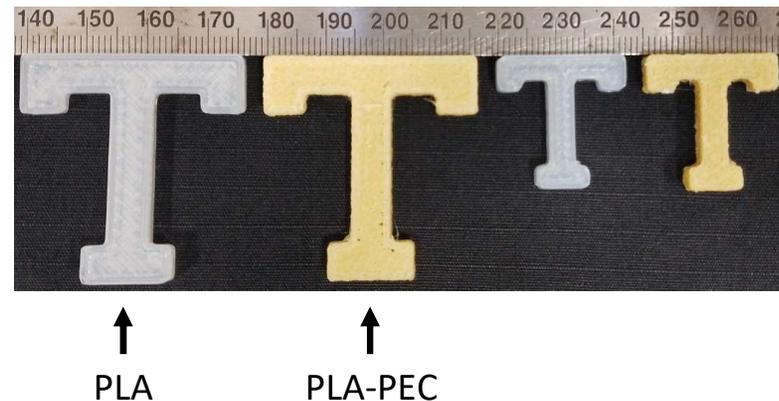
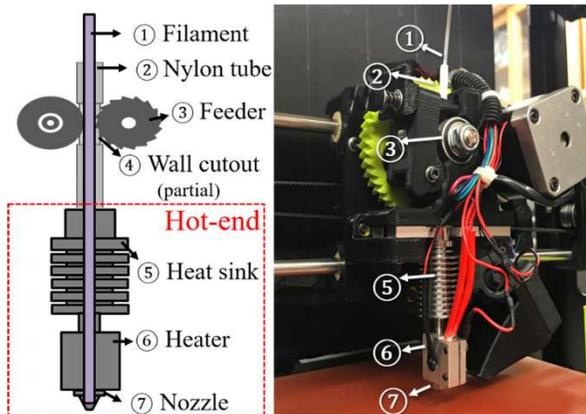




- Filament
 - 25% PEC, 75% PLA
 - Mixed in microcompounder/extruder
 - Plasticized with DI water prior to extrusion
 - Printed at 200 °C, 3000 mm/min
 - Identical to ‘normal’ parameters for PLA



MI-LAB
MANUFACTURING
INNOVATION
LABORATORY



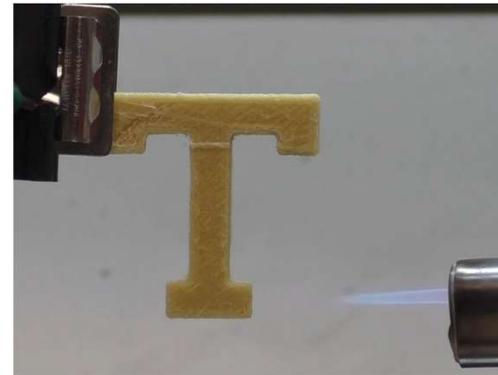
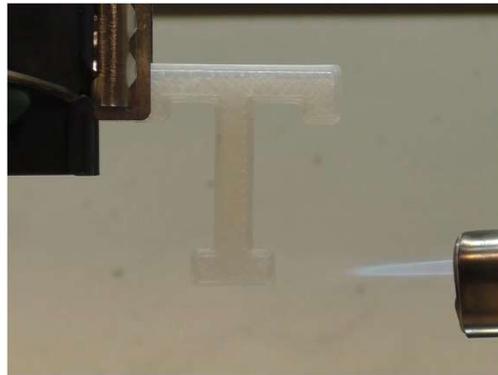
T. J. Kolibaba, C.-C. Shih, S. Lazar, B. L. Tai, J. C. Grunlan, *ACS Materials Letters* 2020, 2, 15.

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- Microscale Combustion Calorimetry

Sample	Char Yield (wt%)	pkHRR (W/g)	pkHRR Temp (°C)	THR (kJ/g)
PLA	0.8 ± 0.2	530 ± 40	392 ± 5	16.8 ± 0.1
PLA-PEC	13.6 ± 0.3	309 ± 3	391	13.6 ± 0.1
Change	+1600%	-42%	-	-19%

- Open flame test



T. J. Kolibaba, C.-C. Shih, S. Lazar, B. L. Tai, J. C. Grunlan, *ACS Materials Letters* **2020**, 2, 15.

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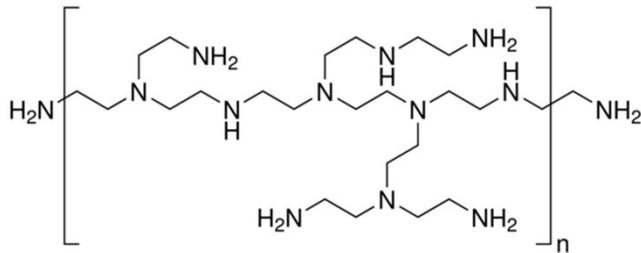
Nanobrick Wall Thermal Shielding Coating

5 wt % Polyethylenimine, Branched (PEI)

Adjusted to pH 6

Sigma-Aldrich

$M_w = 25,000$ g/mol

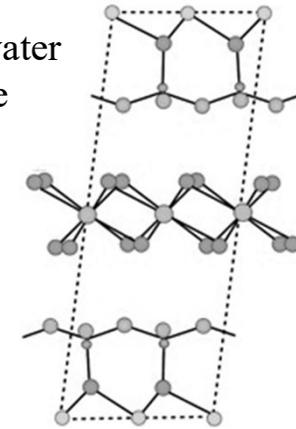


1 wt % Vermiculite Clay (VMT)

Adjusted to pH 10

963++, 7.8 wt% in water

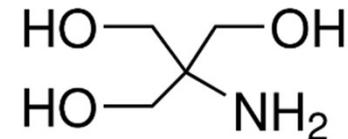
Specialty Vermiculite



50 mM Tris (hydroxymethyl) aminomethane (THAM)

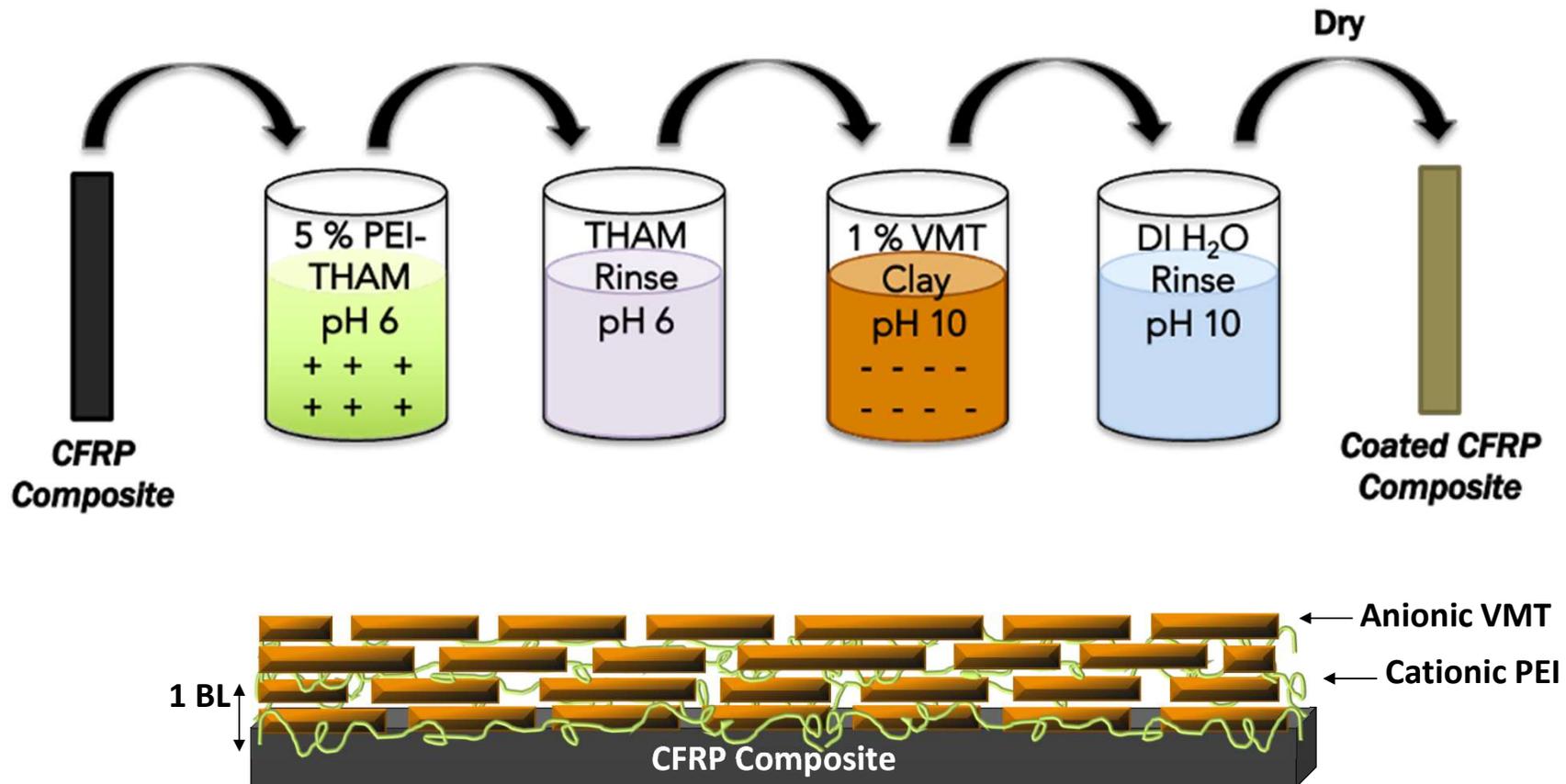
Adjusted to pH 6

Sigma-Aldrich





Deposition of Aqueous Heat Shielding System



3.2 mm Polystyrene Plate



Polystyrene Plate with 8 BL Clay/Polymer Coating (4 mm thick)



- Carbon Fiber Reinforced Polymer
 - 6 BL: 0.1% CH + 50 mM THAM (pH 6) / 1% VMT (pH 10)
 - Dip rinses in 50 mM THAM (pH 6) / pH 10 water
 - Torch test 120 s test with butane blowtorch (2.5 cm away, ~2.5 cm flame)
 - Plate approx. 7.6 x 10 x 0.16 cm³

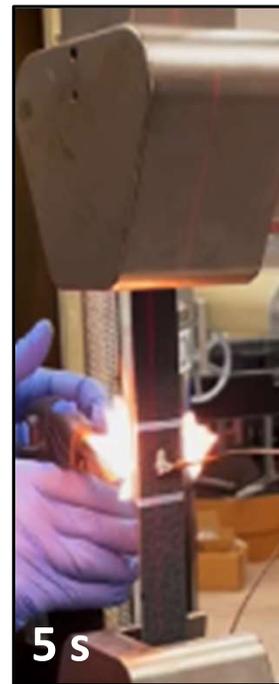
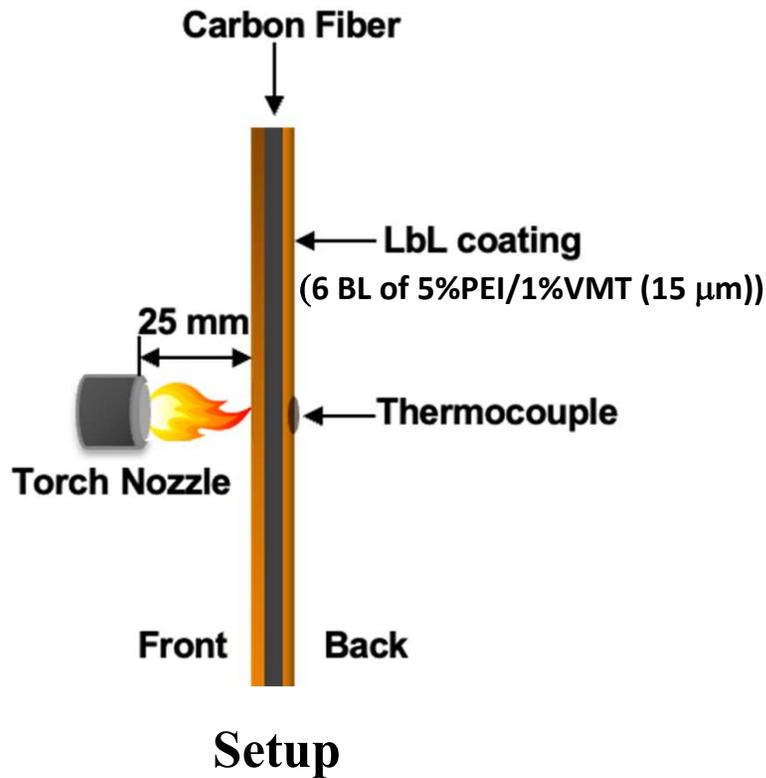


CFRP	% Mass Loss	Backside* Temp (°C)
Uncoated	4.8	353
Coated	2.0	257

*Backside temp is maximum temperature achieved by a thermocouple in contact with the backside of the substrate during the test



Tensile Testing with Torch Exposure



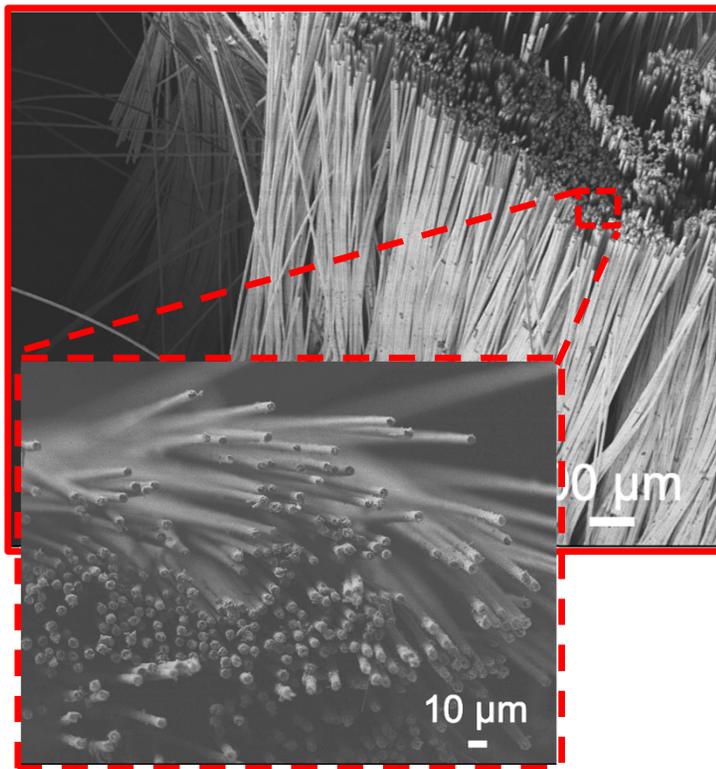
Uncoated



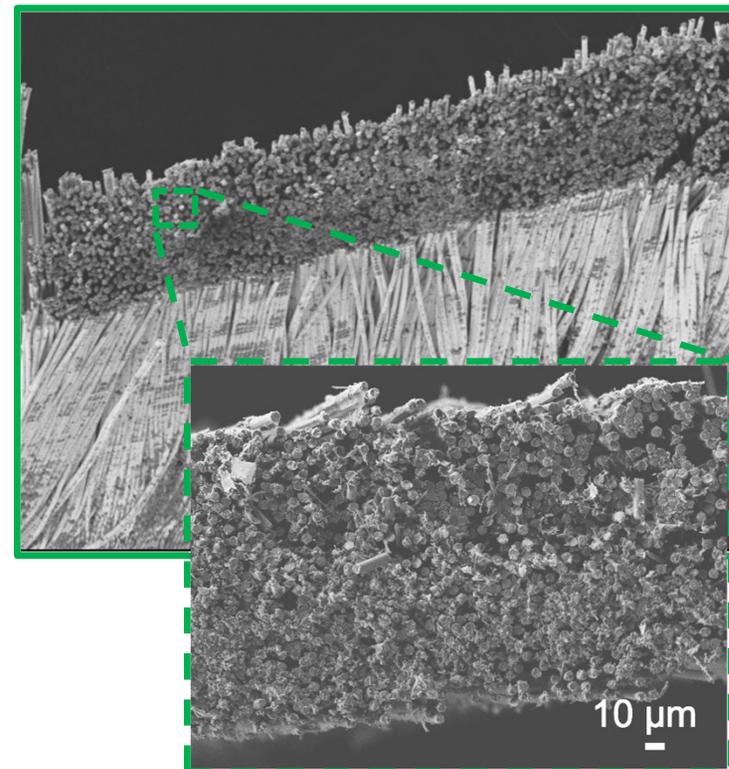
Coated



Uncoated

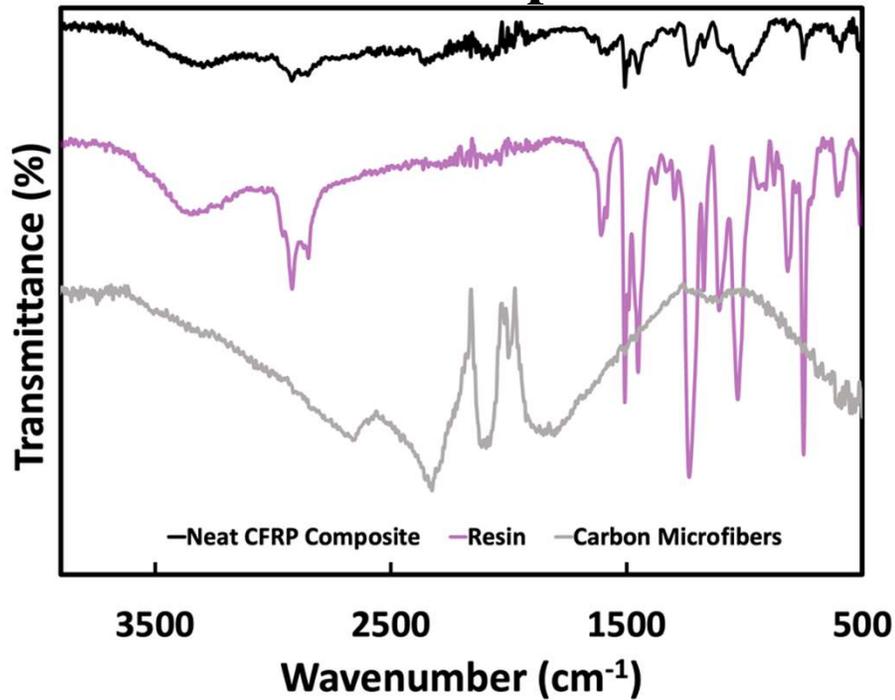


Coated (coating removed)

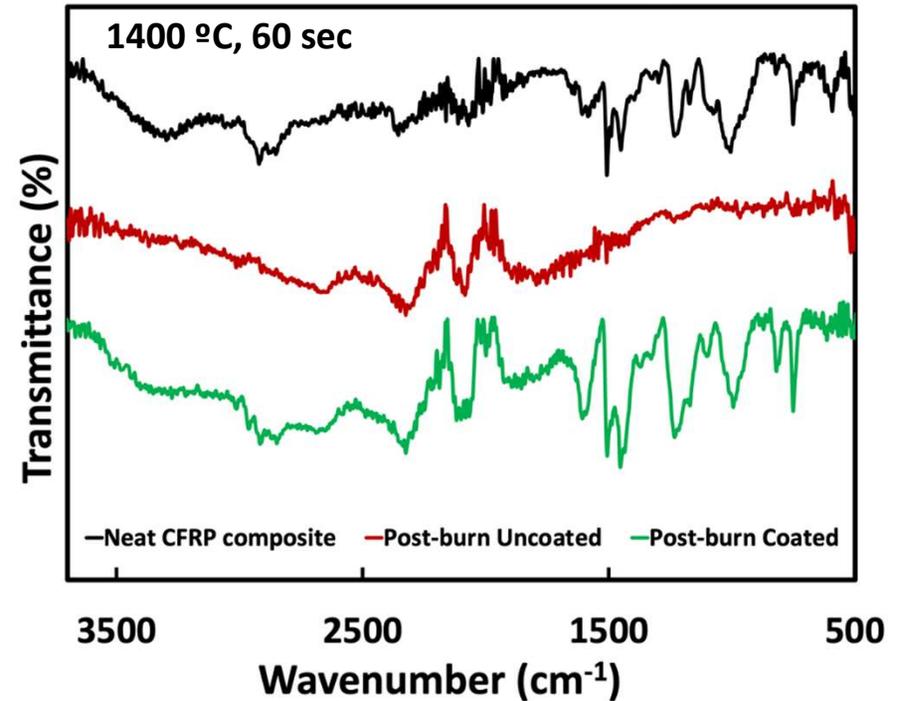




CFRP Components

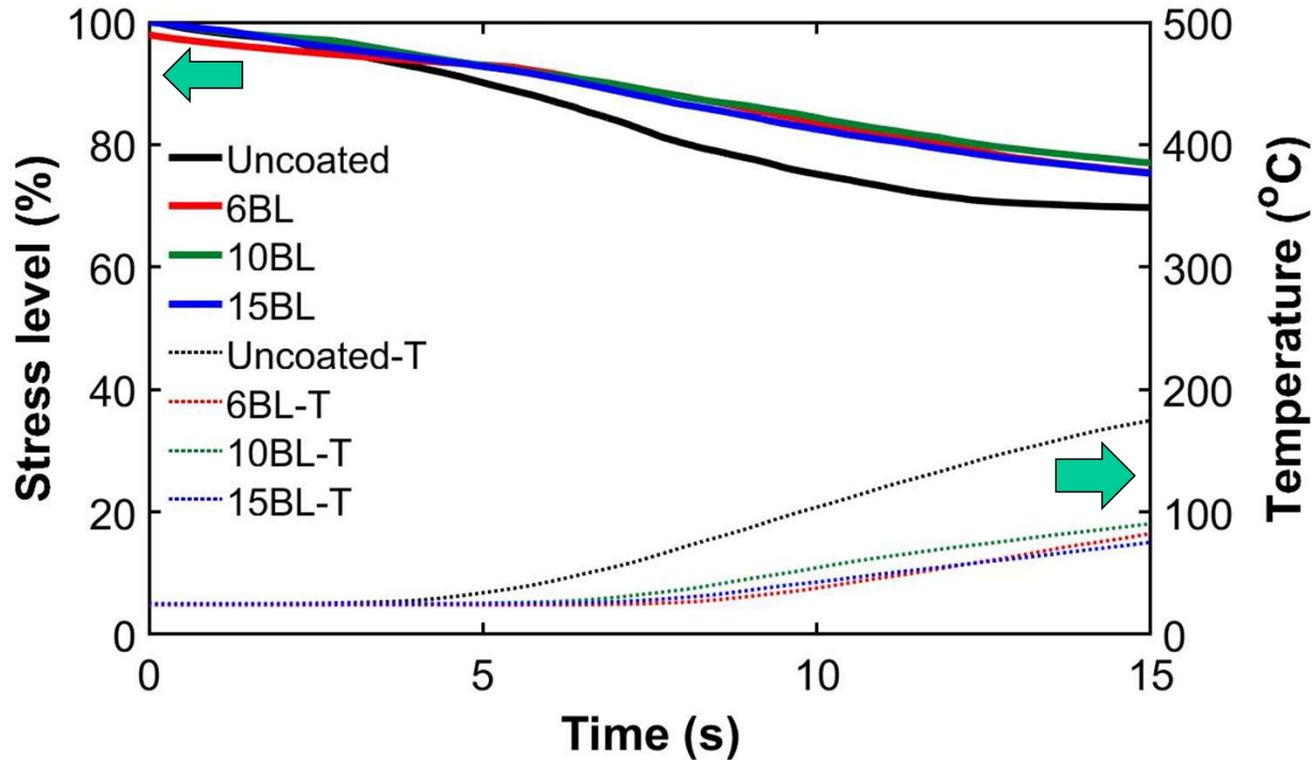


Post-burn CFRP Composites



Epoxy matrix preserved after 1-minute exposure to 1400 °C flame.

Maintaining Strength at High Temperature



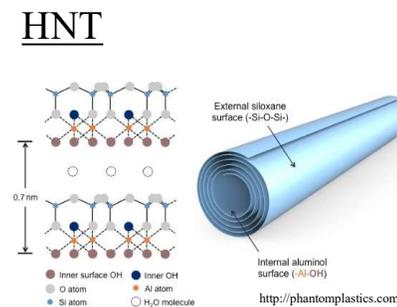
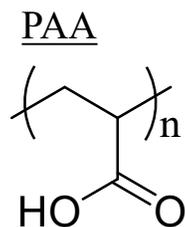
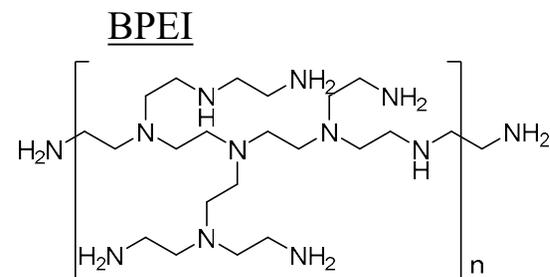
Composite feels lower temperature and maintains strength for longer time upon flame exposure.

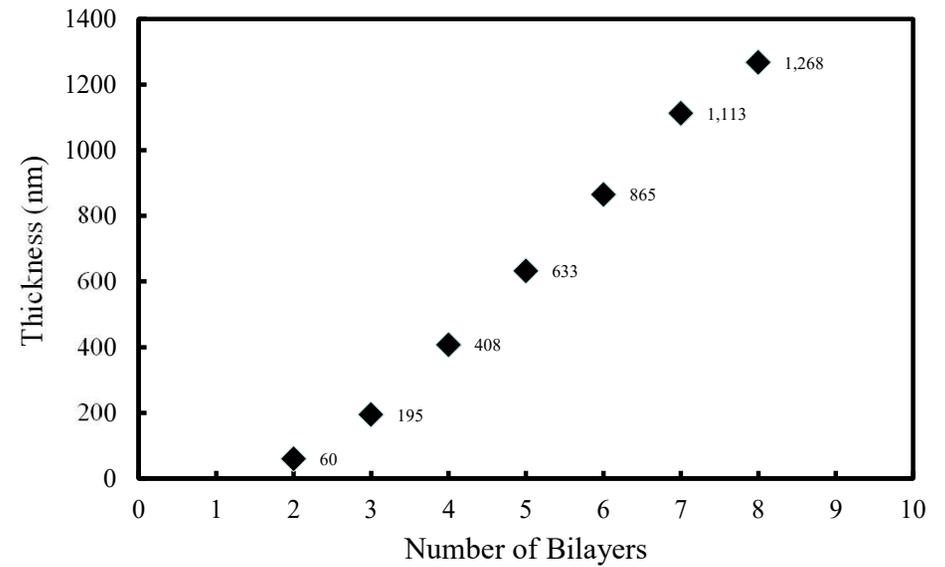
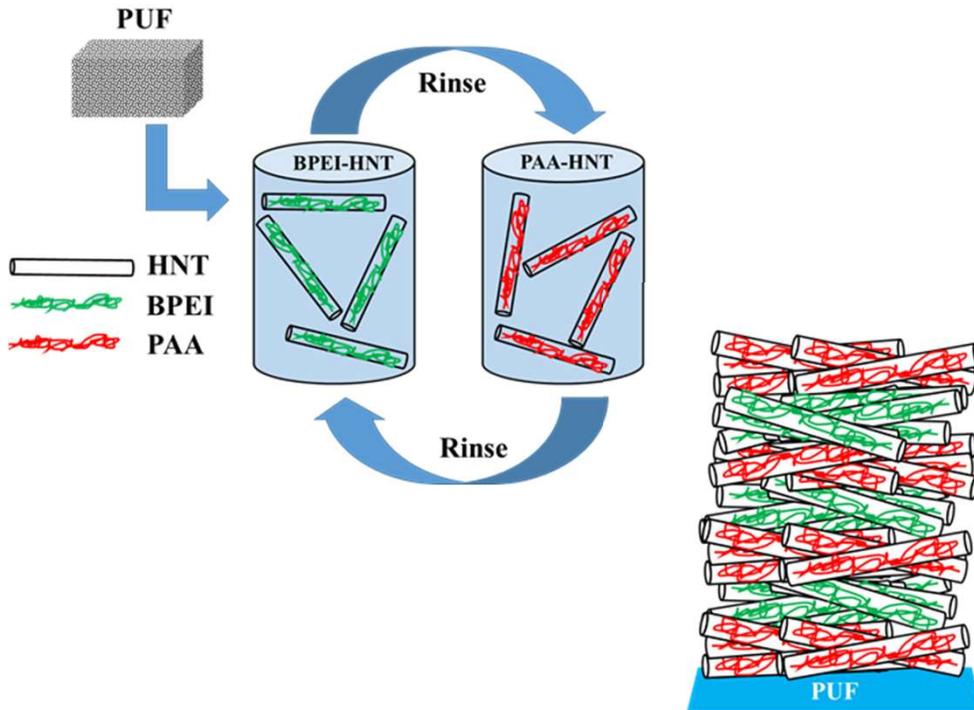
Sample	Atomic Ratio (%)		O / C ratio (%)
	C	O	
Pre-burn uncoated	89.56	10.55	12
Post-burn uncoated	81.80	11.98	15
Post-burn coated*	89.15	10.27	12

*Coating was removed

Post-flame XPS analysis suggests the coated composites did not oxidize.

- **Branched Polyethylenimine (BPEI)**
 - 0.1 wt% in water
 - Purchased from Sigma Aldrich
- **Poly(acrylic acid) (PAA)**
 - 0.1 wt% in water
 - Purchased from Sigma Aldrich
- **Halloysite (HNT)**
 - 0.5% in BPEI and PAA solutions, unaltered pH
 - Ultrasonication to achieve stable suspensions
 - Applied Minerals Inc.



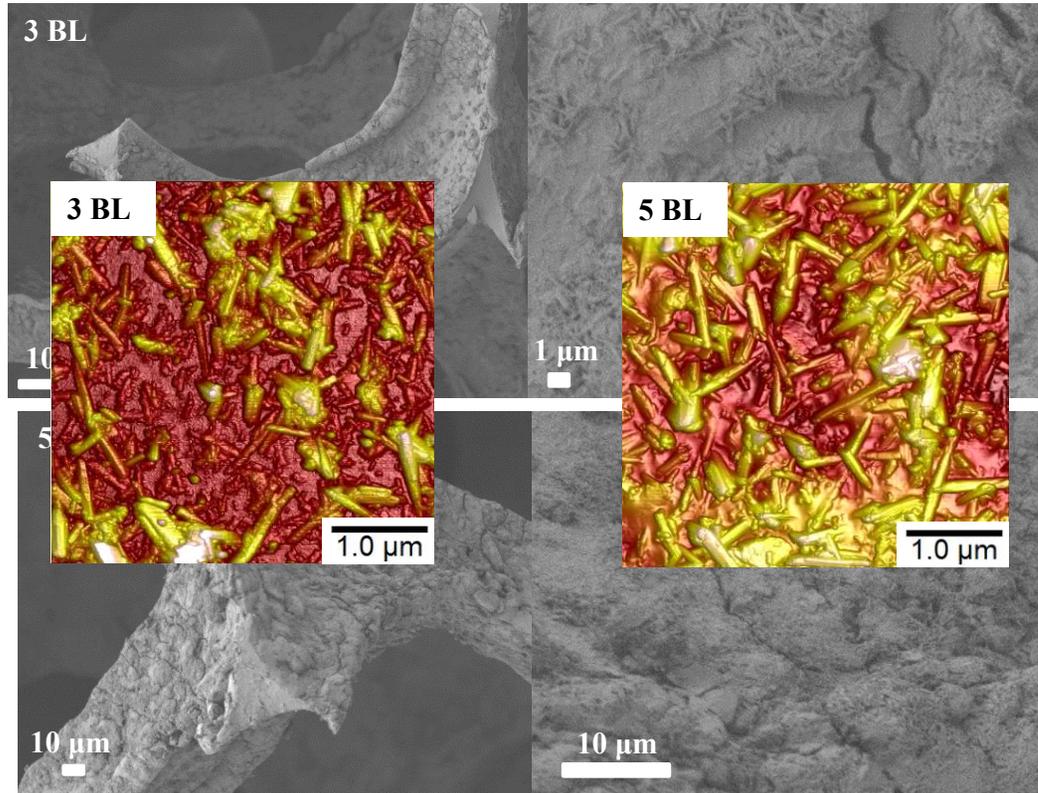
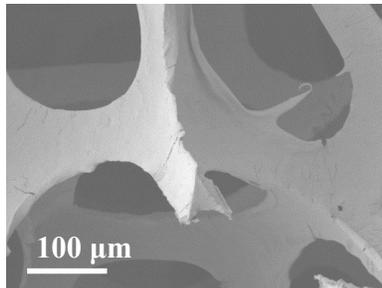


Linear growth observed: 3 BL ~200 nm and 5 BL ~600 nm.

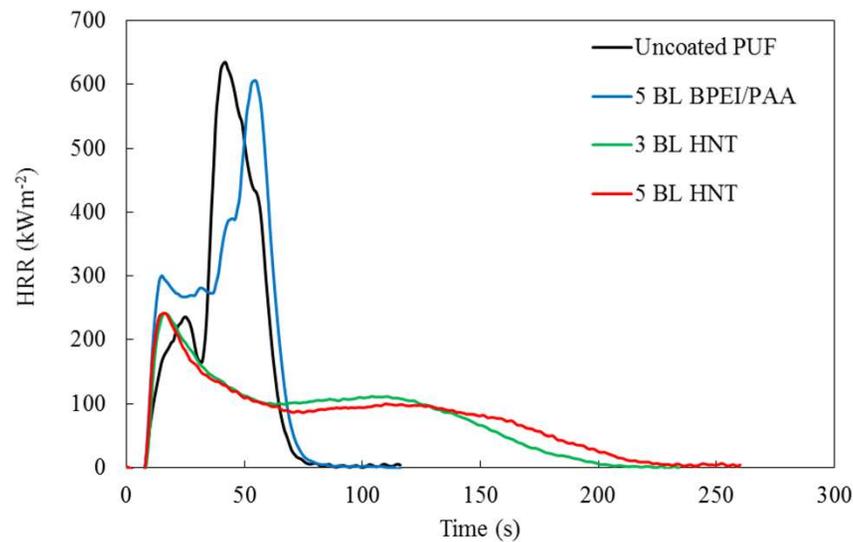


Conformal Coating of Open-Celled Foam

5 BL BPEI/PAA



Uniform coating with HNT nanotube bundles observed in SEM and AFM.



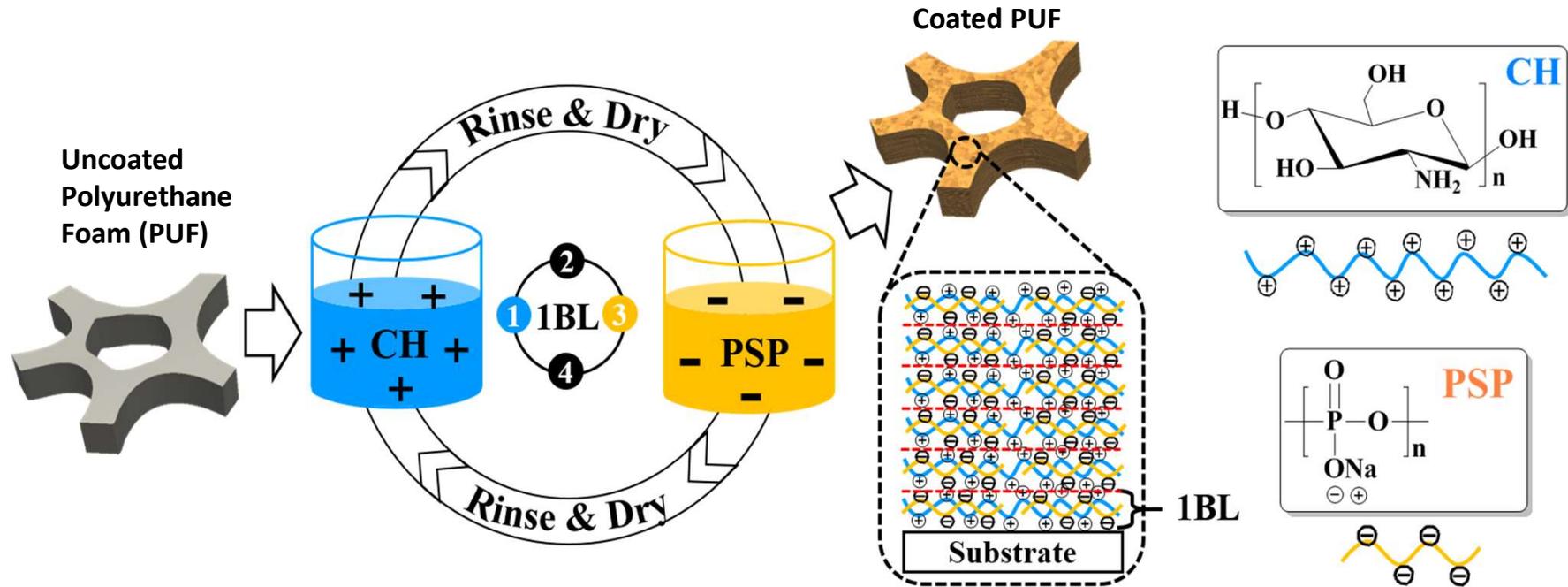
Halloysite reduces flammability

- pkHRR 61% reduction
- TSR 60% at 5 bilayers
- Prevents melt pool formation

Coating	Weight Gain [%]	HNT [%]	pkHRR [kWm ⁻²]	THR [MJm ⁻²]	TSR [m ² m ⁻²]
Uncoated	N/A	N/A	634 ± 31	18.4 ± 0.1	178 ± 7
5 BL PEI/PAA	10.6 ± 0.5	N/A	621 ± 11	20.2 ± 0.3	217 ± 4
3 BL HNT	26.2 ± 0.6	91	244 ± 2	18.1 ± 0.2	93 ± 8
5 BL HNT	34.2 ± 0.5	86	243 ± 2	18.8 ± 0.2	71 ± 7

5BL BPEI-HNT/PAA-HNT
5x Speed

Self-Extinguishing Polyurethane Foam



- Completely polymeric intumescent nanocoating
- CH and PSP both at pH 6
- PSP promotes dehydration of CH to form protective char layer

**Uncoated
Polyurethane Foam**



6 BL CH/PSP

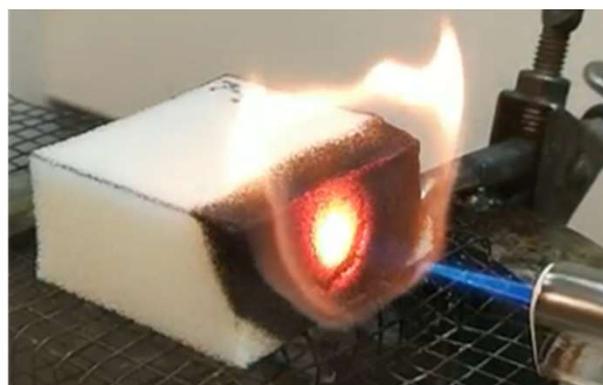


Self-Extinguishes and Maintains Form w/o Clay

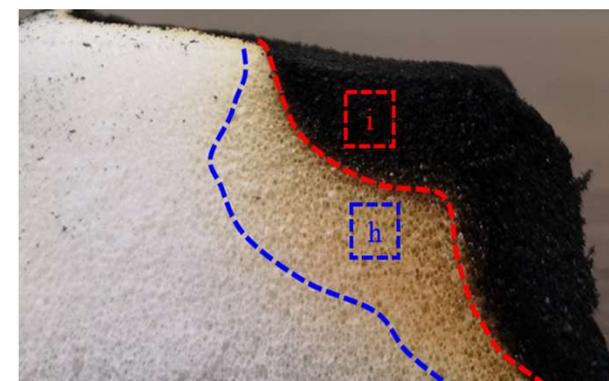
Uncoated



6 BL CH/PSP

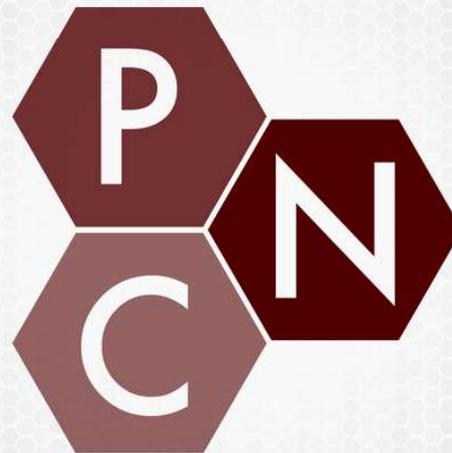


Post-burn 6 BL CH/PSP
cross-section



This clay-free nanocoating exhibits self extinguishing behavior after being exposed to a 1400°C direct flame for 10 s, while preserving structure of the foam.

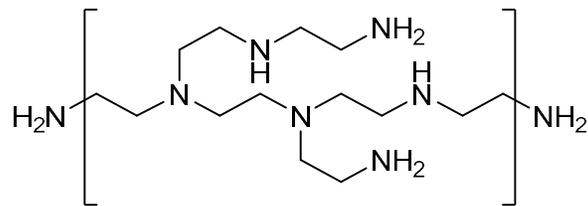
- ✕ Polyelectrolyte complexes (PEC) provide an opportunity to quickly deposit “LbL-like” films
- ✕ Effective flame-retardant coatings deposited in 1 or 2 steps from water-based solution
- ✕ LbL films with few bilayers can yield tremendous properties



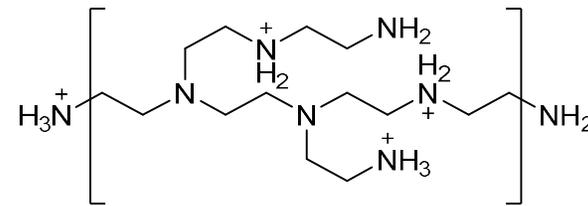
Polymer NanoComposites (PNC) Lab (<http://nanocomposites.tamu.edu>)



- Polyelectrolyte with pH-dependent charge
 - Example: polyethylenimine

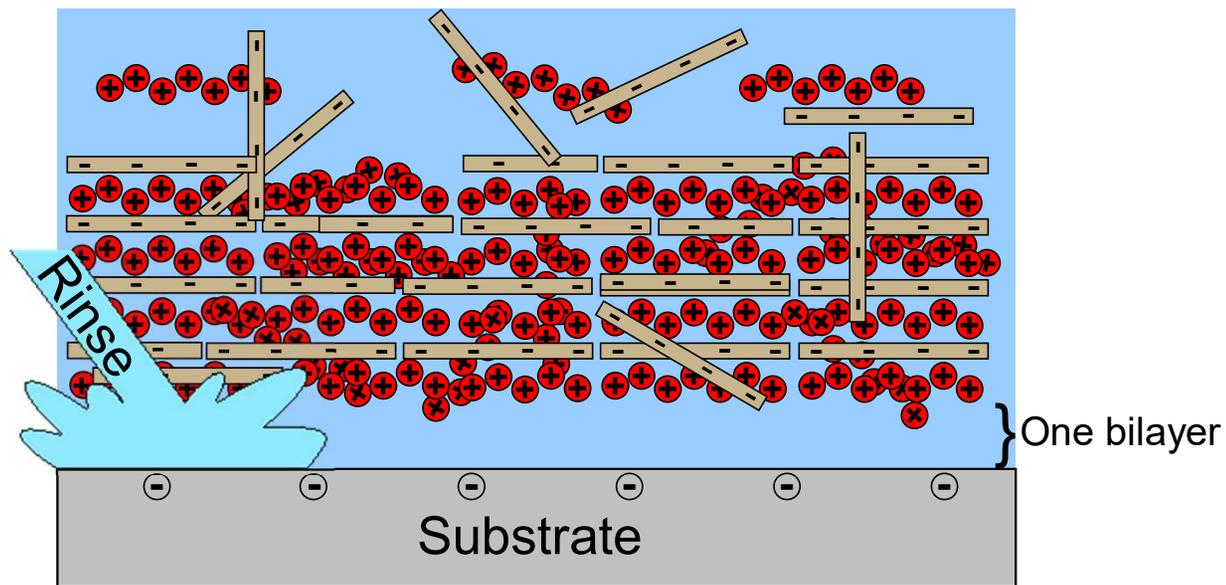


High pK_a
High pH = Low Charge



High pK_a
Low pH = High Charge

- Solution pH can serve as a stimulus to form a PEC on demand



Ambient Processing \neq **Tunable Properties** \neq **Nanoscale Control**

Bertrand, P., Jonas, A., Laschewsky, A., Legras, R. *Macromol. Rapid Comm.* **2000**, *21*, 319.

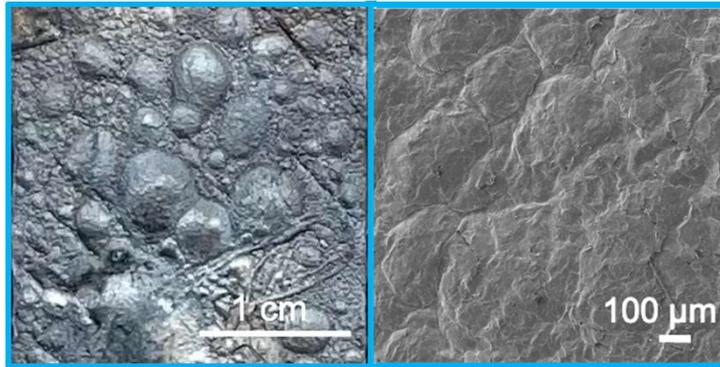
Multilayer Thin Films: Sequential Assembly of Nanocomposite Materials, 2nd Ed., Decher, G., Schlenoff, J. B., Eds., Wiley: New York **2012**.

Ariga, K., Yamauchi, Y., Ryzek, G., Ji, Q. M., Yonamine, Y., Wu, K. C. W. Hill, J. P. *Chemistry Letters* **2014**, *43*, 36.

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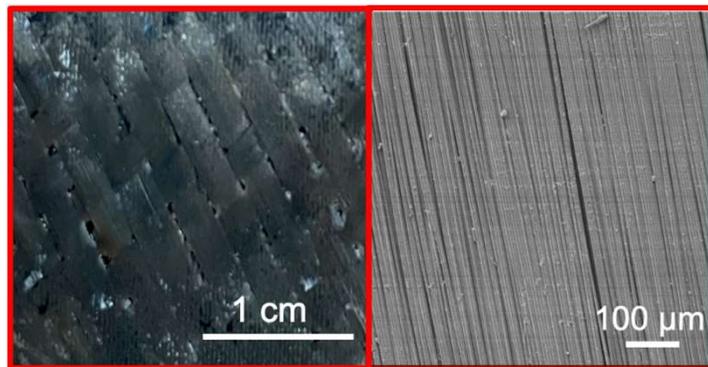


Coated



All images are following one minute exposure to butane torch flame.

Uncoated



Coated (coating removed)

