Detailed Characterization of Emissions from Battery Fires

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Outline

- Background why is this topic important?
- Research overview
- Experimental methods
- Results what we learned
 - Physical observations, gaseous and particle emissions
- Summary important takeaways



Background

- Several battery fire incidents over the last few months
- During failure events, particulates and gases are released
 - Emissions may be toxic, flammable, explosive and could impact visibility in an enclosed space
 - These emissions pose a serious risk to human health either via inhalation or skin absorption
- Critically important to understand composition of particulates and gases emitted from such fires
 - To equip first responders with appropriate PPE
 - To understand impact on people in the vicinity
 - Environmental impact air and water quality
 - To develop mitigation and control strategies





Research Overview

- Objective was to investigate emissions from Li-ion battery fires triggered by thermal runaway
 - Cell level studies focus was on gaseous emissions
 - Included LMO and NCA chemistries
 - Module-level studies focus was on 'fine' particle (<2.5 μ m) emissions and some gases
 - Included LFP and NMC chemistries
- Test programs were designed to gain information on variability, impact of battery chemistry and initiation mechanism on emissions
- For cell level tests, overcharge was used as the abuse mechanism
- For module level tests, the following tests were conducted
 - Test I LFP via nail penetration –
 - Test 2 LFP via nail penetration
 - Test 3 LFP via overcharging
 - Test 4 LFP via overcharging
 - Test 5 NMC via nail penetration

Variability/Repeatability

Variability/Repeatability

Initiation mechanism

Battery chemistry

All test articles were charged to full SOC



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Experimental Methods



Modules were instrumented with temperature and voltage sensors

Sampling location





Cell-level

LMO



NMC



Sample zone

LFP





- Test article placed inside the enclosure
- Particle/gaseous emissions sampled from inception to completion – no control systems were engaged
- Sufficient oxygen was always present to sustain combustion

Emissions Instrumentation



- Focus was on 'fine' particle emissions
 - Sub 2.5 µm particles
- Particle measurements included
 - Particulate Matter mass (PM2.5)
 - Regulated air quality metric
 - Real-time black carbon (or soot) concentration
 - Real-time total particle number/size
 - Includes volatile + solid particles
 - 5.6 nm to 560 nm
 - Real-time solid particle number/size
 - Includes metallic + soot particles (no volatiles)
 - 5.6 nm to 560 nm
- Gas measurements were conducted
 - using an FTIR
 - CO, CO₂, NO, NO₂, HCN, HCI, HF, CH₂O, CH₄ and C₃H₈

Battery Module Instrumentation





- LFP module was instrumented with 16 thermocouples and 6 voltage sensors
- NMC module was instrumented with 16 thermocouples and 5 voltage sensors



NMC module thermocouples

Side 1





NMC module Voltage sensors

Side 2: (-) Terminals



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Results – Cell-Level Tests (Brief)



- Thermal runaway was induced via overcharging
- Test setup simulates an enclosed space with some level of ventilation
- Pretty high levels of formaldehyde were observed for both chemistries
 - Highly toxic compound
- High levels of variability were observed for four tests conducted using similar abuse factors and identical NCA cells



Module-Level Results-1 Physical Observations













NMC nail penetration



- LFP nail-penetration tests
 - Only cells in the path of the nail experienced thermal runaway
- LFP overcharge tests
 - All cells in the module experienced thermal runaway
 - Significant smoke and fire was observed
- NMC nail-penetration tests
 - All cells in the module experienced thermal runaway
 - Thermal runway propagation was observed cell-to-cell
 - Significant smoke and fire was observed

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Module-Level Results-2 Battery Parameters



All modules were charged to full state-of-charge

LFP modules entered thermal runaway after about 15 minutes of overcharging



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Module-Level Results-3 Gaseous Emissions



• LFP via nail-penetration (no significant emissions)





- High emissions observed for multiple gases
- HF exceeded immediately dangerous to life or health (IDLH) limit of 30

ppm



- CO₂ peak 20 times higher than LFP
- Formaldehyde above IDLH limit of 20
 ppm
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11

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Module-Level Results-4 Particle Emissions

Solid PN

emissions,

part./hr

1.56E+15

1.12E+14

8.89E+16

6.11E+16

1.06E+17

Black carbon emissions



Test duration.

sec

260

266

1376

1392

1535

PM2.5

emissions,

g/hr

1.81

0.00

386.09

375.97

551.03

Black carbon

emissions,

g/hr

0.00

0.00

149.90

185.78

66.52

iotal particle number emissions	
1.80E+08	
1.60E+08	
1.40E+08	
1.20E+08	
1.00E+08	
8.00E+07	
6.00E+07	
4.00E+07	
2.00E+07	
0.00E+00	
-2.00E+07 0 200 400 600 800 1000 1200 1400 1600 1800	1
Time, sec	
—Test 1_LFP nail-pen —Test 2_LFP nail-pen —Test 3_LFP OC	
— Test 4_LFP OC — Test 5_NMC nail-pen	

Total nauticle number emissions





Significant PM2.5, soot and particle number emissions were observed for the LFP overcharge and NMC nail penetration tests

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Test

Test 1 LFP nail-pen

Test 2 LFP nail-pen

Test 3 LFP OC

Test 4 LFP OC

Test 5 NMC nail-pen



Total PN

emissions,

part./hr

4.24E+15

1.61E+15

1.13E+17

1.83E+17

2.08E+17

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Module-Level Results-5 Particle Size Distributions



Solid particle size distribution

- Particles were observed to be in the respirable size range
- Peak concentrations were observed to be in the ultrafine particle size range (sub 100 nm) – known to be more toxic to human health
- All five tests exhibited unique size signatures, both, for solid and total particles



Summary

- Emissions from battery thermal runaway events can result in significant particle and gaseous emissions
 - PM emissions includes ultrafine particles that are more harmful to human health (sub 100 nm) in addition to micron sized particles
- Battery chemistry coupled with initiation mechanism influences magnitude of emissions, along with release profile
- Physical dimensions/arrangement of cells within a module could influence the severity of the runaway event
- Emissions from thermal runaway events of identical modules induced into runaway via the same mechanism could be highly variable

 Additional research is required to examine impact of variables such as cell chemistries/formats, SOC, initiation method, packaging, etc.



Thank you for your attention!



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