

I. Group Contribution Method to Predict the Flame Resistance of Polycarbonates

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II. Transparent OSU-Compliant Polycarbonate Copolymers

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the challenge:

BOEING: Flame Resistant Transparent Dust Covers



Ignitability, Melting/Dripping

60-sec vertical
Bunsen burner

6" burn length

15-s specimen
extinguishing time

3-s drip
extinguishing time

Heat release

OSU 65/65: Ohio
State U calorimeter

**65 kW • min/m² total
(during first 2 min)**

**65 kW/m² peak rate
(during first 5 min)**

Smoke release

National Bureau
of Standards smoke
chamber

specific optical
density < 200
(during 4-min test)

Toxicity (OEM):
CO, HCN, HF, HCl,
SO₂, NO_x

expensive, kgs of material,
not "screening" tool

gms of material; many, many
formulations to examine



the solution: convergence of 3 key “events”

1

GE Global Research
GE Plastics (Sabic Innovative Plastics)

POLYCARBONATES



2

“Properties of Polymers:
Their Correlation with Chemical
Structure; Their Numerical
Estimation and Prediction from
Additive Group Contributions”

by DW van Krevelen

3

“Calculating Polymer Flammability from
Molar Group Contributions”
DOT/FAA/AR-01/31
by Richard Walters and Richard E Lyon

..... and a key testing methodology and database (PCFC)

4

Microscale
Combustion
Calorimetry



Flammability Testing (Lyon-Walters)

Flame tests



Bench-Scale



Component-Scale



Milligram

Gram

Kilogram

Metric Ton

Model

Pre-screen

PCFC

Screen

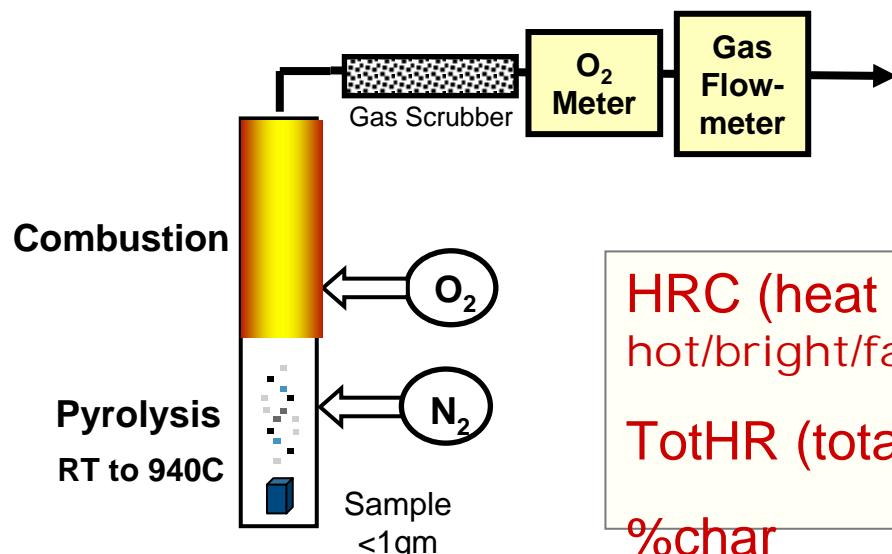
OSU

Large Scale Tests

Small amounts of material
No geometry, processing dependence
Complete combustion, thermodynamic and kinetic effects

5

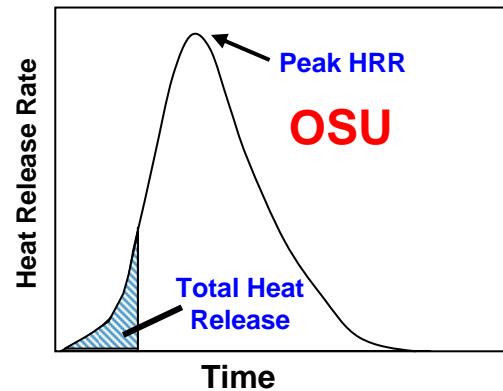
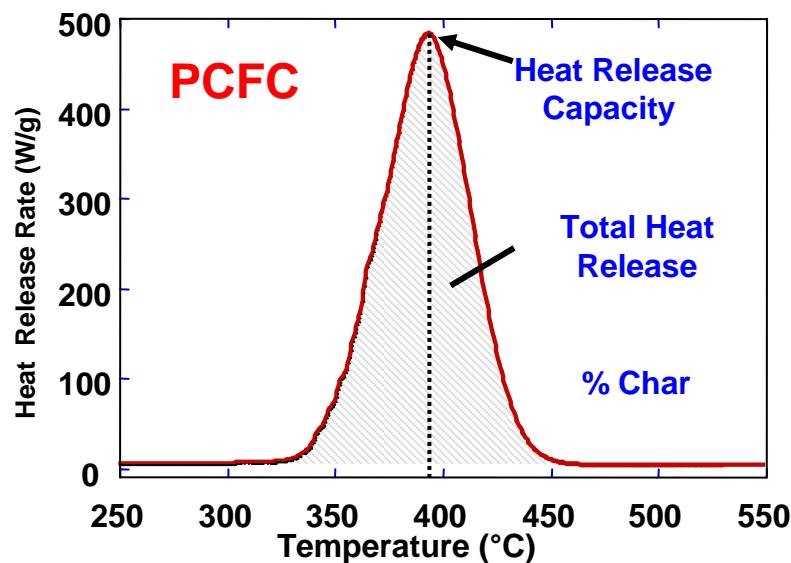
PCFC: Pyrolysis-Combustion Flow Calorimetry (Lyon, Walters)



HRC (heat release capacity) how hot/bright/fast

TotHR (total heat release) how much fuel

%char how much isn't fuel



Key Requirements to Successful Modeling

TEST/DATABASE: LARGE, CONSISTENT, VERIFIED

One operator

Reference materials

One class of
materials

One machine

Literature verification

One test protocol

Polycarbonates

MODEL: SIMPLE, PHYSICALLY APPEALING, PARAMETERIZED

Additivity

Inverse Additivity

Maxwell Model

OPTIMIZATION: model to data

~90% consistent (~10% outliers)

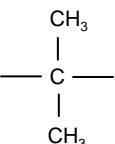
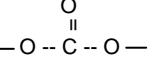
User Beware: Don't look too hard,
you're bound to find "inconsistencies"

Cannot predict special effects,
transitions, geometric or processing
artifacts (drip, buckle, melt, skin
formation, bubbling....)



Example: additivity of group contributions

BPA-polycarbonate

						
Group Contribution to HRC	13.1	85.5	13.1	13.6	$\Sigma=125.3$	$(\frac{kJ}{mole-^{\circ}K})$
Molar Mass	76	42	76	60	$M=254$	$(\frac{gm}{mole})$

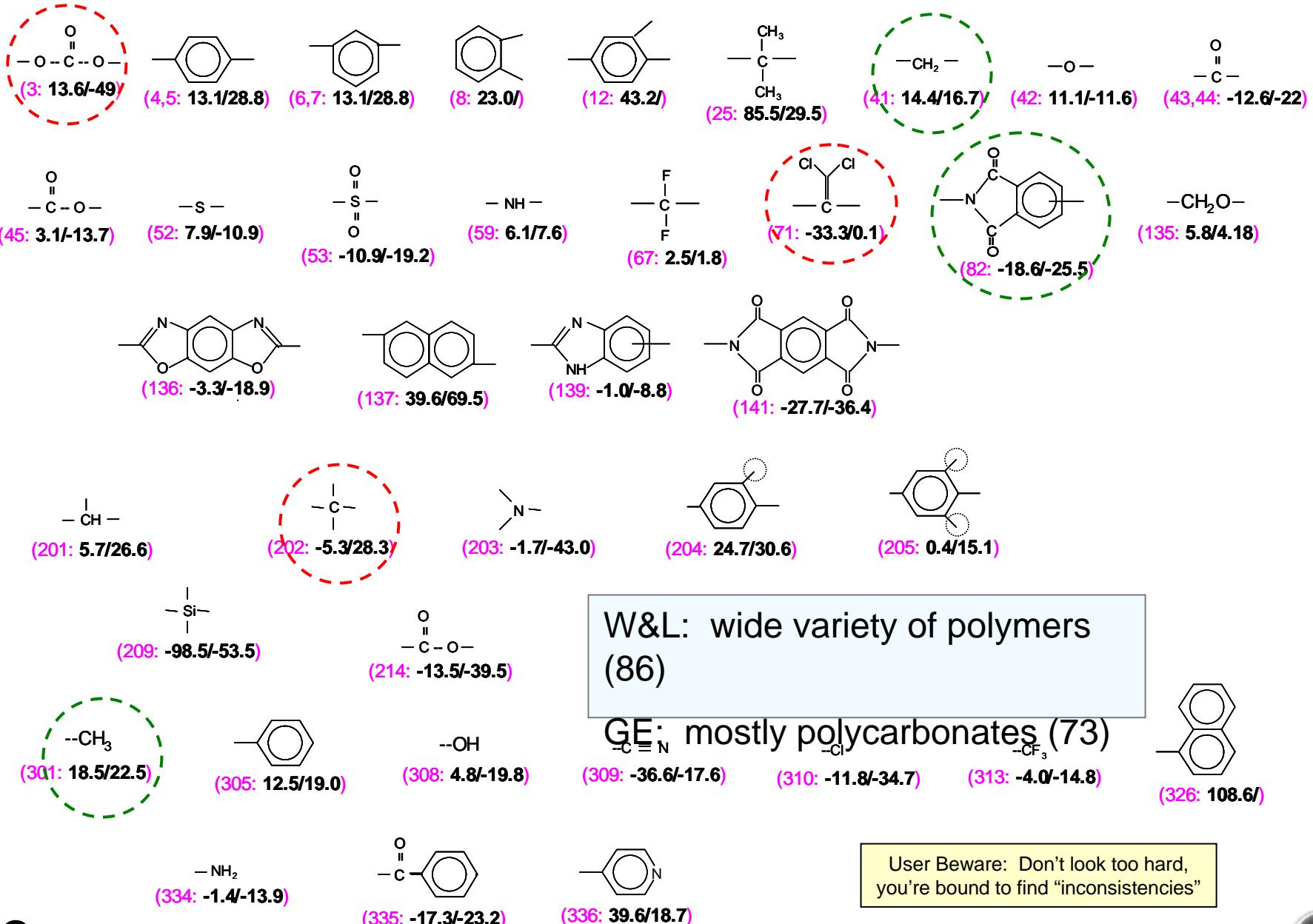
$$HRC = \Sigma/M = 492 \text{ J/g-}^{\circ}\text{K}$$

(some measured values: 528, 476, 504, 547)

HRC: Optimization

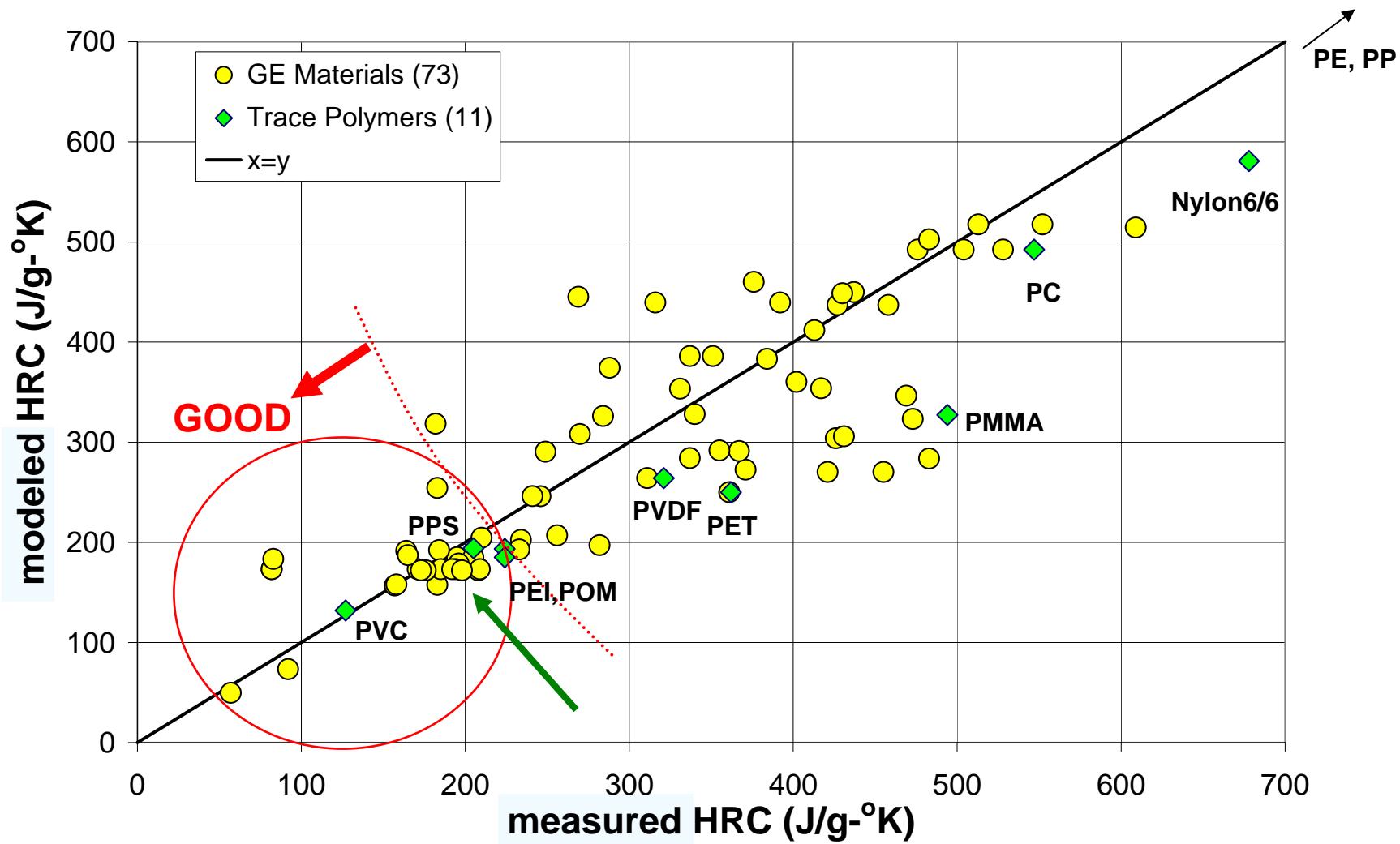
Ψ_i (kJ/mole/°C)
model/W&L

38 groups (W&L had 42)



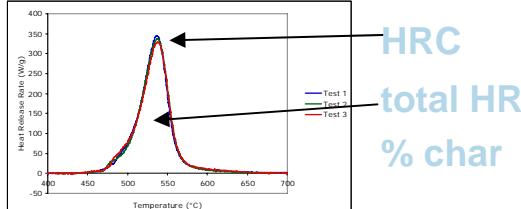
Optimization of HRC data

Fit to 84 tests



“how brightly does it burn”

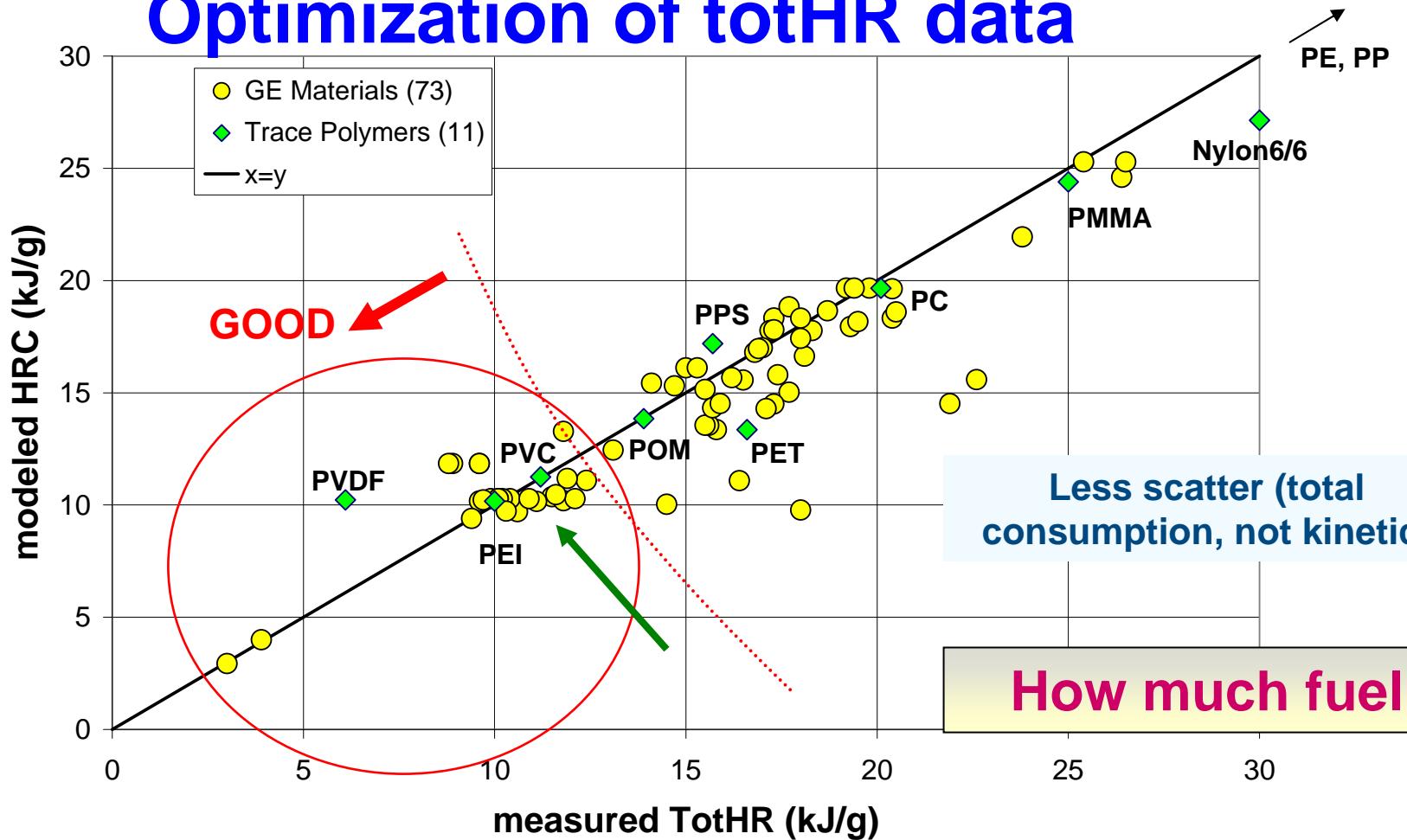
HRC < 200 J/g/°K



Expanded PCFC modeling

3 Variables, 3 Models, 3 Criteria

Optimization of totHR data

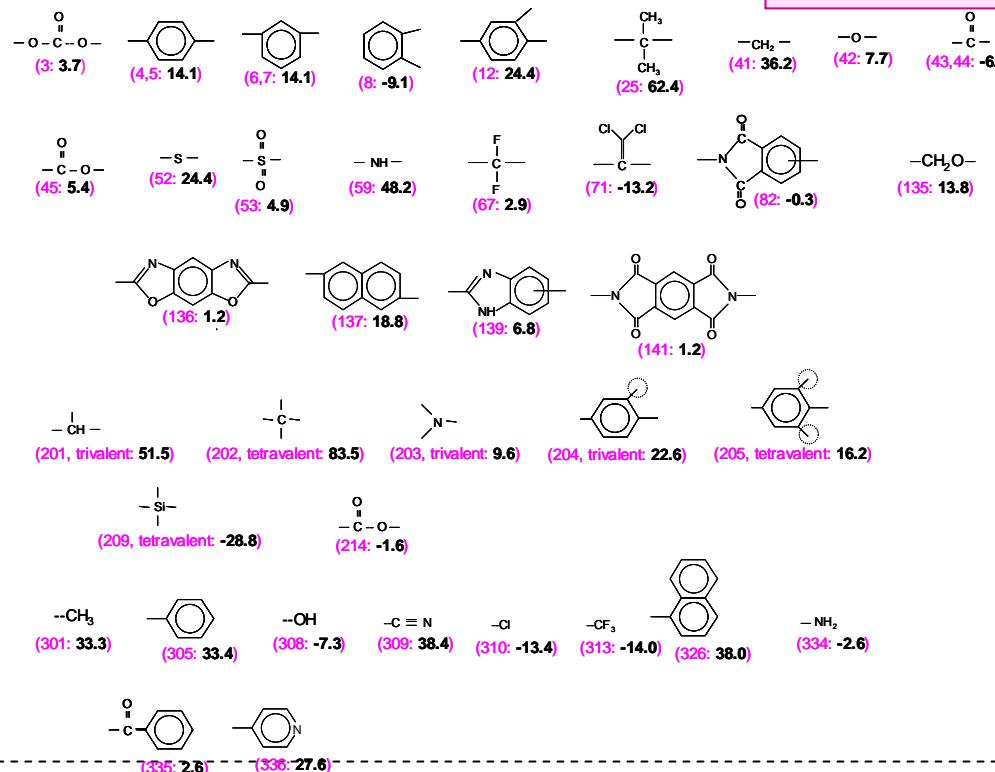


FR Criterion: Total Heat Release < 12kJ/g

Tot HR: Optimization

kJ/gm

Model for Total Heat Release



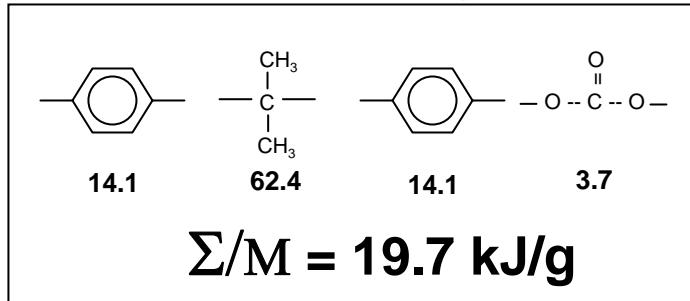
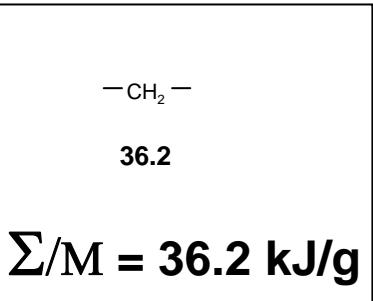
We have normalized (divided) each group's contribution by their respective molar mass, to illustrate their "total fuel content"

Each identified "group" has a particular "contribution" that is "added" into the sum Σ

PE

examples

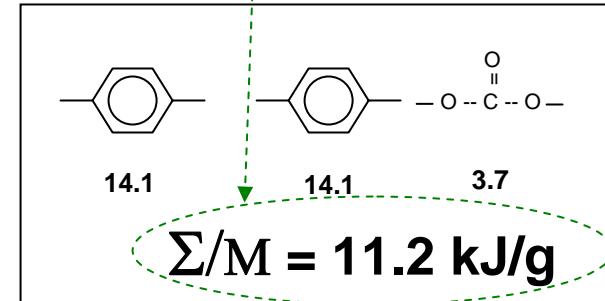
BPA-PC



high "fuel content"

medium "fuel content"

BP



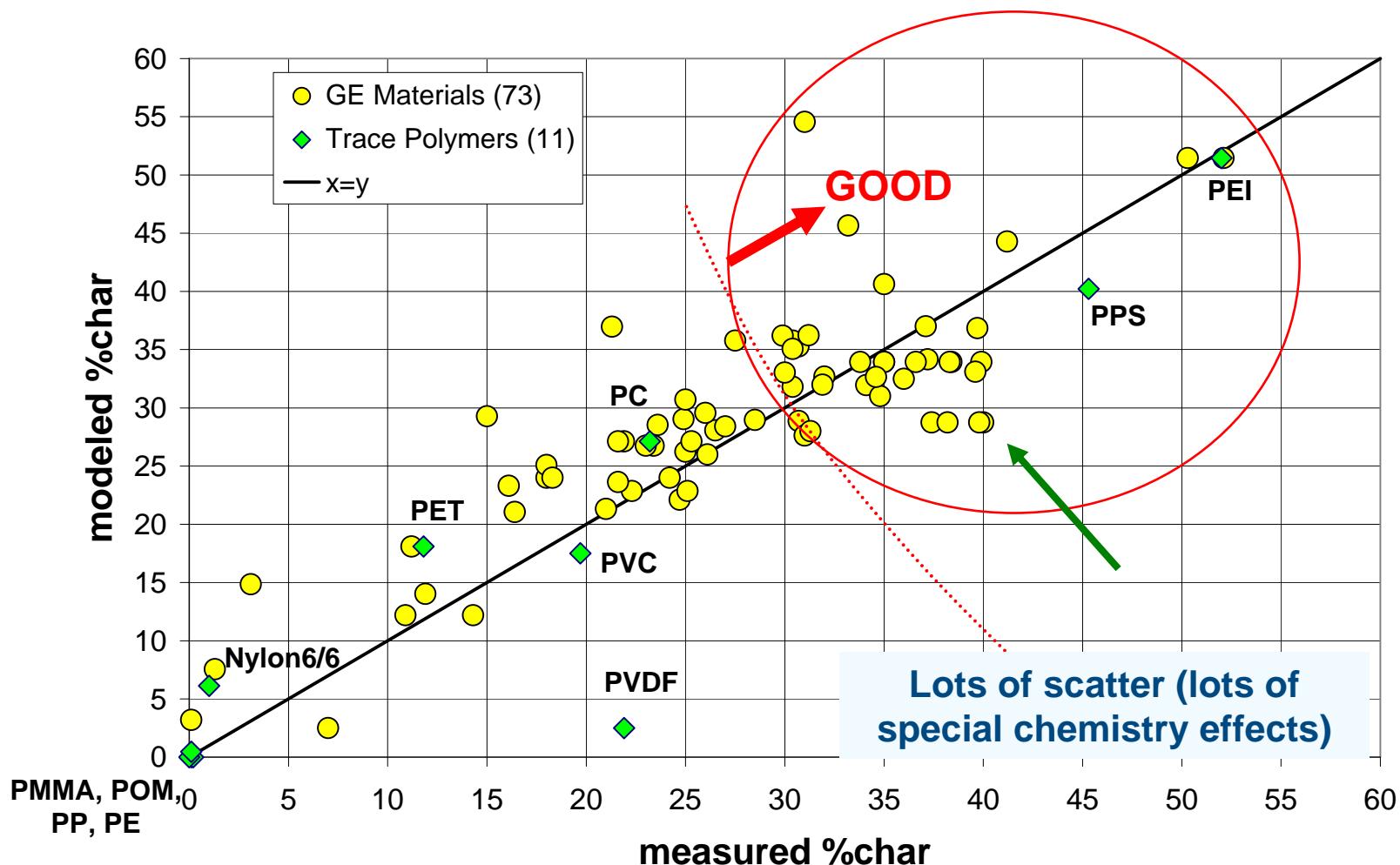
low "fuel content"

imagination at work

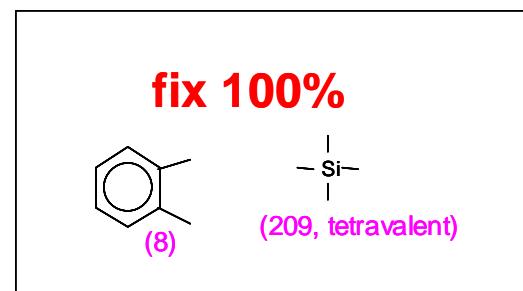
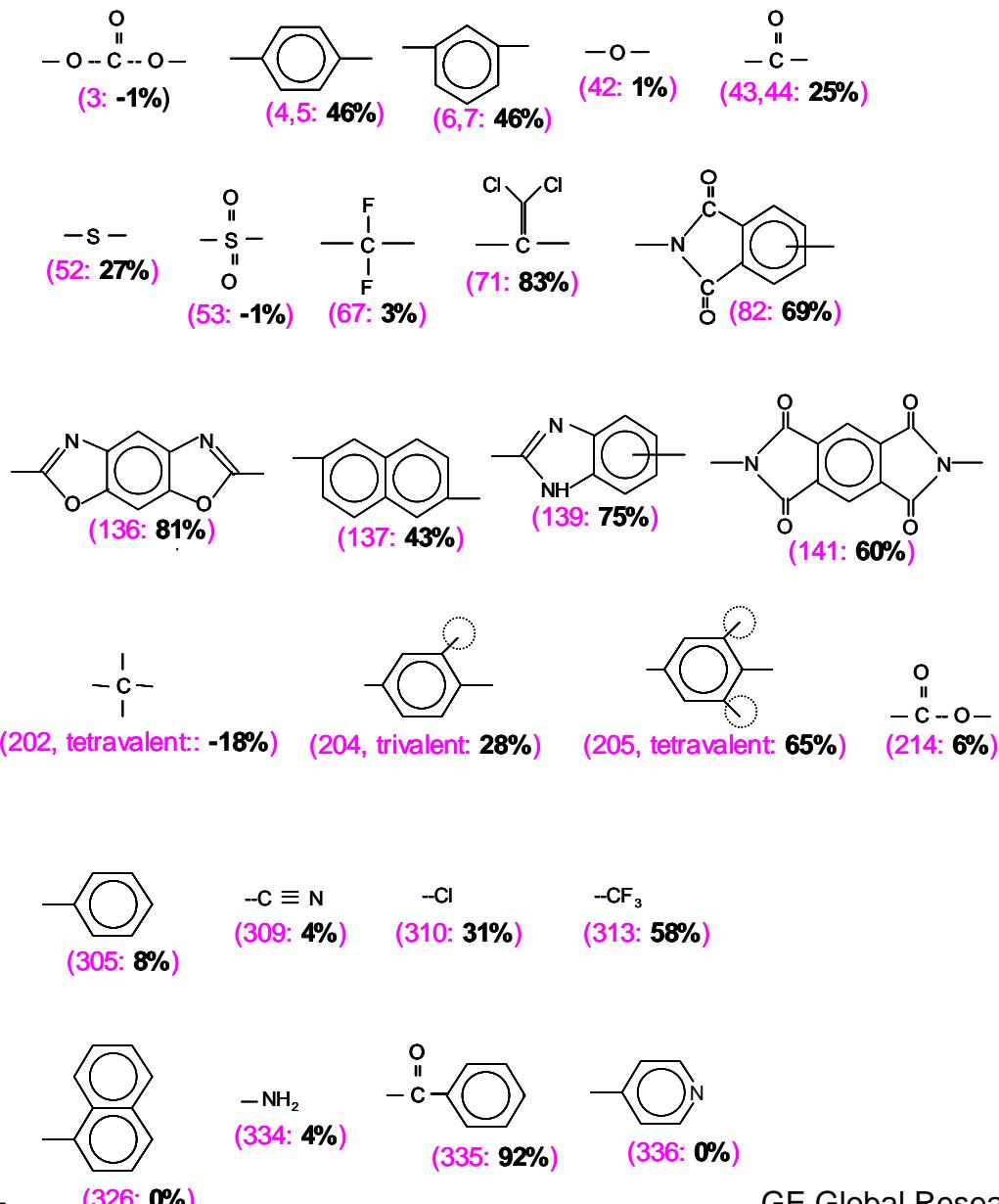


Similar Modeling for % char

How much isn't fuel?



Char: Optimization



Summary

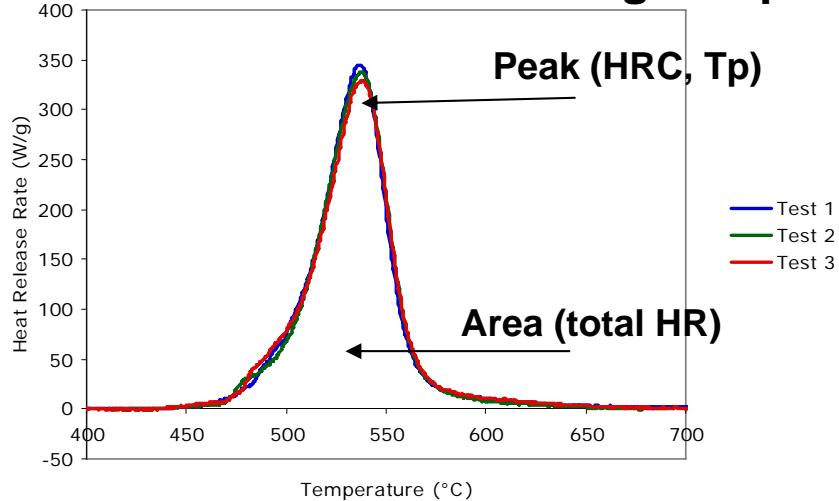
- Lyon-Walters HRC model specialized for polycarbonates using PCFC data
- New additive group contribution model for total heat release and % char developed
- Established 3 FR Criteria: $HRC < 200 \text{ J/g/oK}$; $\text{totHR} < 12 \text{ kJ/g}$; $\% \text{char} > 30\%$
- Successful prediction of several FR transparent formulations

2 More Things

- PCFC testing vs OSU testing
- An even simpler “atomic contribution” model??

PCFC vs. OSU Tests: Correlations?

PCFC: Measured on mg sample



Total HR = **how much will burn**

%Char = how much will not burn

HRC = **how brightly will it burn**

totHR < 12 kJ/g

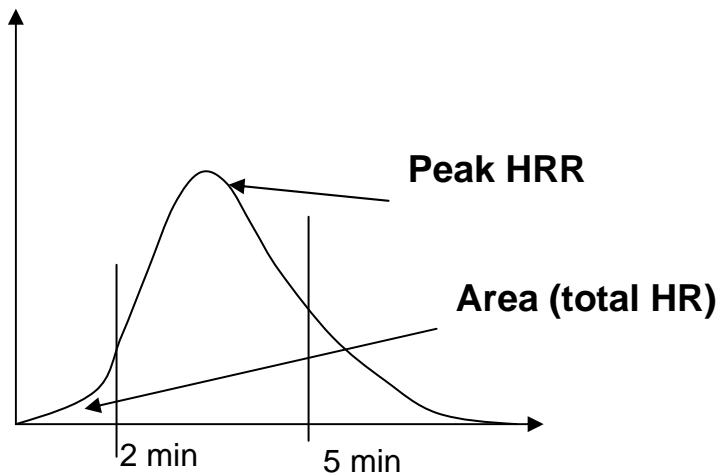
HRC < 200 J/g^o

K

OSU = f (kinetics, thermodynamics, rheology, geometry, drip, bubbles....)

PCFC – total combustion/flammability
OSU – additional factors important

OSU: Measured on a large part



How much will burn in 2 mins

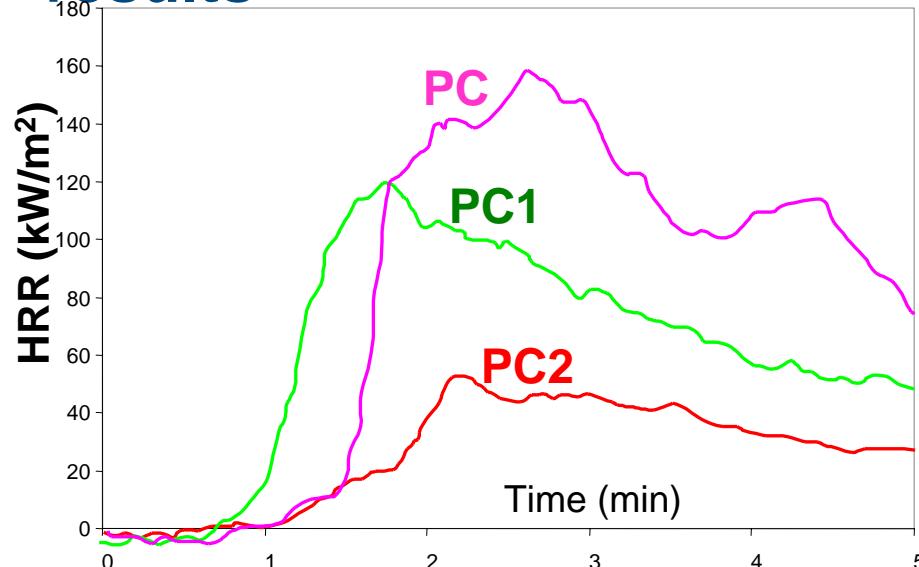
How brightly will it burn in 5 mins

totHR (2min) < 65 kW-min/m²

Peak HRR (5min) < 65 kW/m²

FR agents in PC: OSU results differ from PCFC

OSU
results

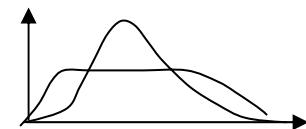


PC1>PC (2min totHR~same; peak low)
PC2>PC

PCFC

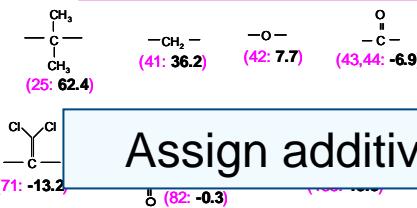
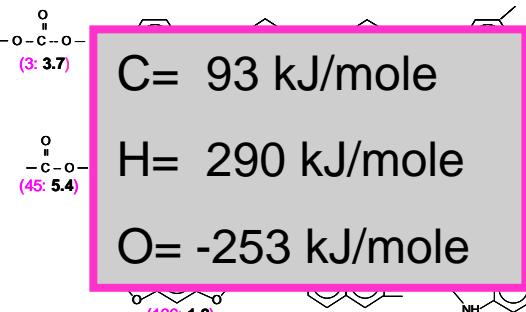
Composition	HRC	Total HR	Tp	% Char
PC	476	19.4	557	25.3
PC2	134	9.3	491	24.1
PC1	453	19.3	541	23.8

PC1~PC
PC2>PC

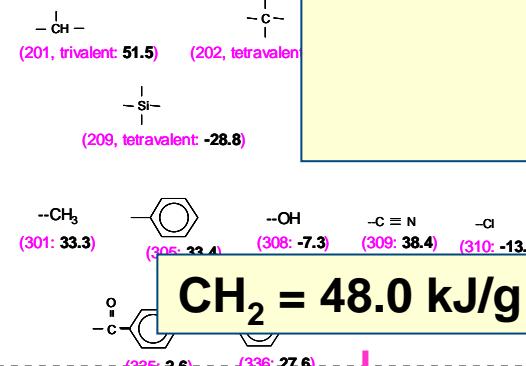


PCFC test captures the reduction of heat release for PC2;
Does not capture the time-dependence behavior of PC &
PC1

Atomic Model for Total Heat Release



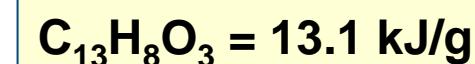
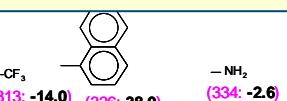
Assign additive group contribution value to atoms



$$\begin{aligned}
 \text{C}_{16}\text{H}_{14}\text{O}_3 &= [16(93) + 14(290) + 3(-253)]/[16(12) + 14(1) + 3(16)] \\
 &= [1488 + 4063 - 758]/[192 + 14 + 48] \\
 &= 4792/254 = \mathbf{18.8 \text{ kJ/g}}
 \end{aligned}$$

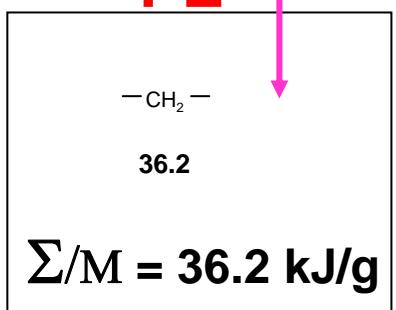
has a
' that is

“added” into the sum Σ



PE

examples



high “fuel content”

45.3 meas

19.6 meas Research

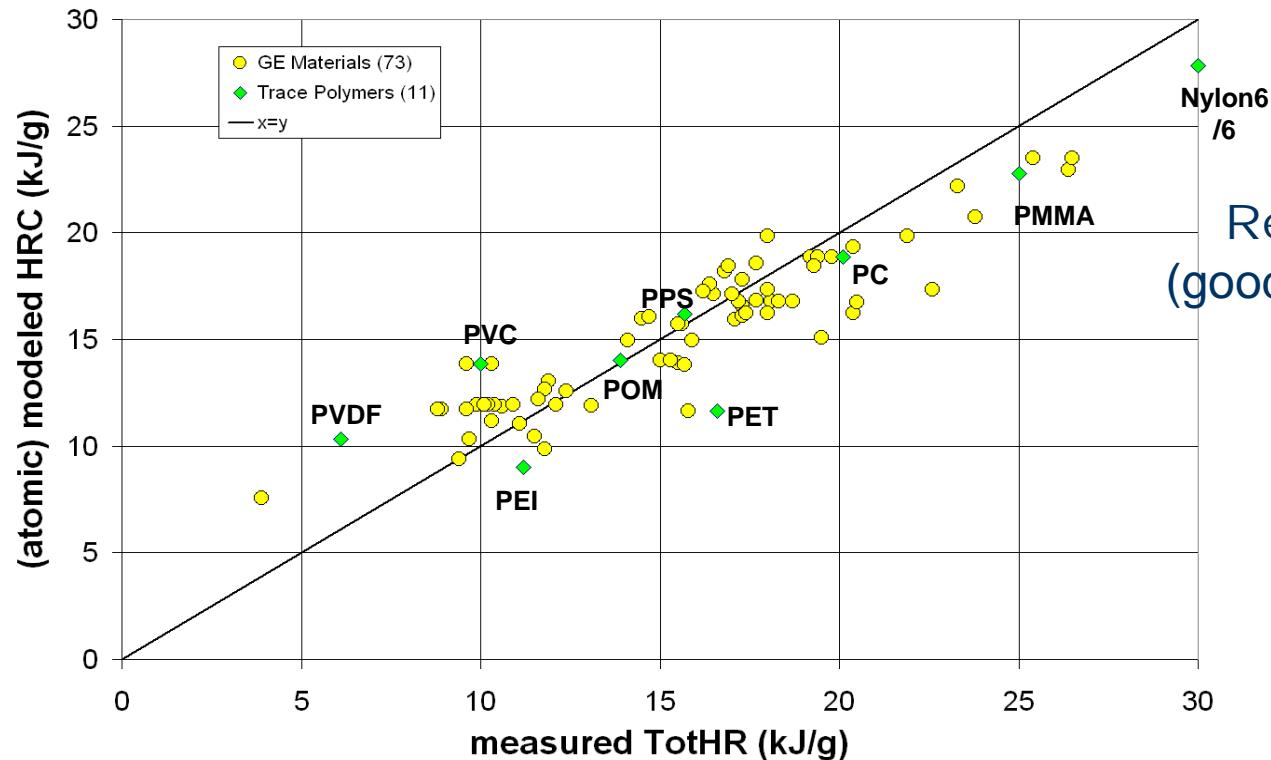
19.6 Measuring Research

low “fuel content”

imagination at work

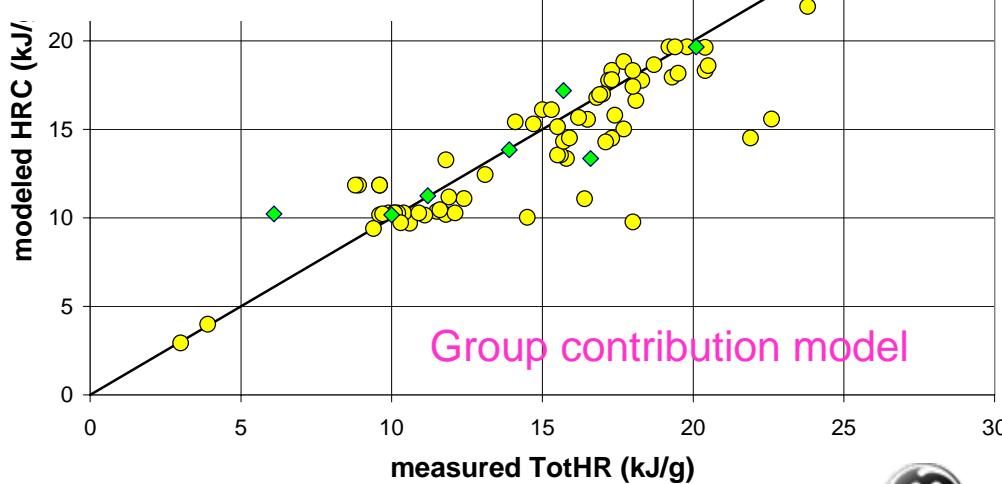


Atomic model for totHR



Result of optimization
(good fit to simple atomic model)

C= 93 kJ/mole
H= 290 kJ/mole
O= -253 kJ/mole
N= -345 kJ/mole



Group contribution model

User Beware: Don't look too hard,
you're bound to find "inconsistencies"

